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1 **Production Strategy of Functional Oligosaccharides from Lignocellulosic**  
2 **Biomass Using Enzymatic Process: A Review**

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8

9 **Abstract** Lignocellulosic biomass was previously known as agricultural waste, and  
10 its only use was in the production of compost. Over time, this waste has evolved  
11 into one of the raw materials used in the production of biofuels or bioethanol, an  
12 alternate substrate for enzyme manufacturing, and so on. Due to the abundance of  
13 lignocellulosic biomass and its potential derivate compounds, scientists have  
14 devised a process from lignocellulose fractions (cellulose and hemicellulose) to  
15 high-value cello-oligosaccharides (COS) and xylo-oligosaccharides (XOS). These  
16 oligosaccharides provide a variety of health benefits, including anti-diabetic and  
17 prebiotic characteristics. However, due to slow reactions and low yields, the  
18 manufacturing of these oligosaccharides remains an issue. As a result, in this paper,  
19 we discuss the possibilities of a multi-step process and the use of thermostable  
20 cellulases for improve COS dan XOS production from lignocellulosic biomass. We  
21 also mention from the potential pretreatment process to the "adsorption-separation"  
22 method for separating  $\beta$ -glucosidase from thermostable cellulases produced for the  
23 manufacture of XOS and COS from lignocellulosic biomass

24

25 **Keywords:** Cellulase, COS, Lignocellulose, Multi-Step, Thermostable, XOS.

26

27 **Overview**

28

29 Lignocellulosic biomass was formerly known as waste from agricultural products  
30 and its utilization was only up to the manufacture of compost. Along with the times,  
31 this waste has developed into one of the raw materials in the manufacture of biofuels  
32 or bioethanol by simultaneous saccharification and fermentation (Itelima et al.  
33 2013). Some study has looked into using lignocellulose biomass (such as rice straw)  
34 as an alternative substrate for the production of thermostable enzymes (Yuansah et

35 al. 2019). Due to the abundance and its potential derivate compounds to explore,  
36 lignocellulosic biomass is attractive to study, especially as a substrate for the  
37 production of high-value cello-oligosaccharides and xylo-oligosaccharides.

38 Lignocellulose is formed by three major biopolymer components: cellulose,  
39 hemicellulose, and lignin (Pointner et al. 2014). Cellulose and hemicellulose are a  
40 carbohydrate that consists of glucose, xylose, mannose, arabinose, and galactose  
41 (Wang et al. 2016). When the homopolysaccharide, cellulose, is hydrolyzed  
42 partially, it will release cello-oligosaccharides and a little amount of glucose. While  
43 heteropolysaccharide, hemicellulose, will release some oligosaccharides (such as  
44 xylo-oligosaccharides (XOS), manno-oligosaccharides (MOS), arabino-xylo-  
45 oligosaccharides (AXOS), etc.) and monosaccharides (Saville and Saville 2018;  
46 Yuansah 2019) . The conversion process from lignocellulosic biomass to high-value  
47 oligosaccharides can be carried through various pretreatment processes and  
48 enzymatic hydrolysis. However, enzymatic conversion of lignocellulosic biomass,  
49 particularly the cellulose fraction, is considered a slow hydrolysis reaction.

50 The proper pretreatment techniques will optimize the oligosaccharides yield  
51 from a lignocellulose complex structure that is highly resistant to enzymatic  
52 hydrolysis (Zoghlami and Paës 2019). Before entering the enzyme hydrolysis  
53 phase, several inhibitors such as lignin and other polyphenolics must be handled in  
54 the pretreatment step (Qin et al. 2016; Zoghlami and Paës 2019). The pretreatment  
55 procedure breaks down the crystalline structure of cellulose and reduces the degree  
56 of polymerization, removing all lignin and breaking down hemicelluloses (Baruah  
57 et al. 2018). Proper pretreatment procedures are necessary to acquire the highest  
58 yield of oligosaccharides from the bioconversion of various lignocellulosic  
59 substrates.

60 For large-scale production of oligosaccharides from lignocellulosic biomass,  
61 enzymatic hydrolysis of lignocellulose biomass at high temperatures utilizing  
62 thermostable enzymes shows exceedingly promising. Using thermophilic or  
63 thermotolerant microbes to produce thermostable enzymes, we can overcome the  
64 limitation of lignocellulose hydrolysis. When a reaction can be carried out at a high  
65 temperature using thermostable enzymes, it is quicker, more efficient, and less  
66 prone to contamination (Giovannoni et al. 2020). As a result of the stability of  
67 thermostable enzymes, more specialized products and fewer by-products are  
68 produced, providing for a longer hydrolysis time and greater flexibility in the

69 configuration process (Giovannoni et al. 2020; Vasconcellos et al. 2015). The  
70 thermostable cellulases used produced by thermostable enzymes in the process  
71 should be separated from the  $\beta$ -glucosidase leaving only the endoglucanase and  
72 exoglucanase acting on the substrate. This multi-step process improved the total  
73 COS yield accumulated in each step by approximately 51.78% with the hydrolysis  
74 of approximately 75.56% (Chu et al. 2014). According to the literature, adequate  
75 pretreatment and a multi-step method using thermostable cellulose optimize  
76 enzymatic activity and product yields of XOS and COS. As a result, this article  
77 begins with a brief description of lignocellulose sources and their potential as  
78 enzyme substrates for functional oligosaccharides production. With the idea that  
79 suitable pretreatment and a multi-step enzymatic process enhanced product yields,  
80 the possible source as enzymatic substrates seeks to make it a little easier for  
81 researchers to find the optimal source for the manufacture of COS and XOS.

82

### 83 **Lignocellulose Source for Functional Oligosaccharides Production**

84

85 <sup>45</sup> Agricultural waste is the primary source of lignocellulose, which is made up of  
86 <sup>7</sup> biopolymers such as cellulose, hemicellulose, and lignin. The agro-waste product  
87 <sup>22</sup> lignocellulose consists of cellulose (40-50%), hemicellulose (20-30%), and lignin  
88 (10-25%) (Maheshwari 2018; S. Sharma et al. 2019). Because of their availability  
89 and abundance of essential polysaccharides, cellulose (consisting of D-glucose) and  
90 hemicellulose (consisting of glucose, mannose, xylose, arabinose, galactose, etc.),  
91 <sup>6</sup> agricultural waste such as rice straw (Z. Hu et al. 2021; A. L. Li et al. 2018), wheat  
92 <sup>7</sup> straw (Collins et al. 2014; Shrivastava et al. 2014), sugarcane bagasse (de Souza et  
93 <sup>42</sup> al. 2013; Guilherme et al. 2015), cacao pod (Lu et al. 2018), sago pith residue  
94 (Husin et al. 2019), corncob (M. Li et al. 2014; Silva et al. 2015), apple pomace  
95 (Hijosa-Valsero et al. 2017), and red algae (Rabemanolontsoa and Saka 2013) can  
96 be potential candidates as an enzymatic substrate to produce xylo- and cello-  
97 oligosaccharides <sup>2</sup>

98 The main component of lignocellulosic biomass is cellulose, which is the  
99 most widespread and easily accessible carbohydrate polymer on the planet and a  
100 primary polysaccharide element of plant cell walls (Maheshwari 2018). Cellulose  
101 is a homopolysaccharide, a carbohydrate biopolymer composed of monomeric  $\beta$ -  
102 <sup>31</sup> D-glucopyranose linked together by a 1,4 glycosidic bond. The structure of the

103 cellulose chain consisted of 500-1400  $\beta$ -D-glucopyranose (D-glucose) molecules  
104 that packed together to form microfibrils. A microfibril (cellulose fibrils) is formed  
105 up of microfibrils that have been joined together. Because of the presence of  
106 cellulose fibrils, lignocellulosic biomass is particularly resistant to enzymatic  
107 hydrolysis (Zoghalmi and Paës 2019). Cello-oligosaccharides can be manufactured  
108 by hydrolyzing cellulose using  $\beta$ -glucosidase deficient cellulases (Chu et al. 2014).  
109 Hemicellulose is the second most common substance after cellulose (Maheshwari  
110 2018). Hemicellulose is a heteropolysaccharide made up of several different  
111 carbohydrate monomers, particularly pentoses sugar units such as xylose and  
112 arabinose, hexoses sugar units such as mannose, glucose, and galactose, along with  
113 acylated sugar which is represented in varying ratios in diverse materials (Dionisi  
114 et al. 2014; Maheshwari 2018). In hardwoods, xylan is the most common  
115 biopolymer, while in softwoods, glucomannan is the most abundant (Dionisi et al.  
116 2014). Hemicellulose can be partially hydrolyzed and degraded into  
117 xylooligosaccharides (XOS) utilizing hemicellulases (Moser et al. 2014).

118

### 119 **Enzyme Inhibitors in Lignocellulosic Biomass**

120

121 The presence of inhibitors is one of the challenges in using lignocellulosic  
122 components as substrates in enzymatic hydrolysis. In lignocellulosic waste,  
123 inhibitors are derived from lignin, phenolic groups, sugars, and their derivatives (B.  
124 Hu et al. 2016; Qin et al. 2016). Inhibitors can form naturally and accumulate in  
125 biomass, or form as a result of a process. The majority of phenolic compound  
126 derivatives inhibit the enzyme activity of amylase and cellulase (Table 1) (Desseaux  
127 et al. 2018; González-Bautista et al. 2017). However, some inhibitors can also act  
128 as enzyme activators for various enzymes. *T. reesei* cellulase enzyme, for example,  
129 is activated by certain concentrations of ferulic, syringic, sinapic, and vanillic acids  
130 (Stamogiannou et al. 2021). Knowing the substance's potential inhibitors allows  
131 you to decide what steps to take to solve the problem. Avoiding materials with few  
132 inhibitors will help the enzymatic process. The lower the inhibitor content of the  
133 substances, the milder the pretreatment required. Nonetheless, the use of resistant  
134 materials is highly desirable to bring about a significant change in waste treatment  
135 (Jönsson et al. 2013).

136

137 **Table 1.** Enzyme Inhibitors in Lignocellulosic Biomass

Inhibitor	Source	Inhibited Enzymes	Ref
Crude Phenolic	Sugarcane Bagasse, apple pomace, pecan	CMCase, xylanase, $\alpha$ -amylase, $\alpha$ -glucosidase	(Feng and Kong 2022; González-Bautista et al. 2017; Hijosa-Valsero et al. 2017)
p-coumaric acid	Corn stover, Cocoa pod husk	Cellulase, xylanase	(Chen et al. 2020; Lu et al. 2018)
Quercetin	Cocoa pod husk, Cocoa powder, <i>Mucuna pruriens</i> seed	Cellulase	(Lu et al. 2018; Qin et al. 2016; Rai et al. 2017; Sorrenti et al. 2020; Stamogiannou et al. 2021)
Kaempferol	Water lily	$\alpha$ -glucosidase, cellulase	(Mugaranja and Kulal 2020; Qin et al. 2016; Stamogiannou et al. 2021)
Tannins / Tannic Acid	Oak Bark, Acorn Caps	$\beta$ -glucosidase, cellobiohydrolase, endoglucanase	(Jönsson et al. 2013; Liu et al. 2021; Majewska et al. 2022; Mhlongo et al. 2015)

Gallic Acid	Douglas fir, Black wattle, Cocoa pod husk, <i>Mucuna pruriens</i> seed	$\beta$ -glucosidase, endo- $\beta$ -1,4- xylanase	(Jönsson et al. <sup>48</sup> 2013; Lu et al. 2018; Mathibe et al. 2020)
Vanillic acid	Douglas fir, Black wattle, vanilla orchid	endo- $\beta$ -1,4-xylanase, $\alpha$ - amylase, $\alpha$ -glucosidase	(Aleixandre et <sup>41</sup> al. 2022; Gallage and Møller 2015; Mathibe et al. 2020)
Vanillin	Cocoa powder, vanilla orchid	Endocellulase, exocellulase (cellobiohydrolase), $\beta$ - glucosidase. Amyloglucosidase, xylanase	(Gallage and <sup>50</sup> Møller 2015; Hidayatullah et al. 2020; Sorrenti et al. 2020)
Syringic acid	Grass hay, Oak woodchip	$\alpha$ -amylase, $\alpha$ -glucosidase <sup>43</sup>	(Aleixandre et al. 2022; Ziolkowska et al. 2020)
Syringaldehyde	Grass hay, Oak woodchip	Endocellulase, exocellulase (cellobiohydrolase), $\beta$ - glucosidase	(Ziolkowska et al. 2020)
Trans-cinnamic acid	<i>Cinnamomum cassia</i> bark (Chinese cinnamon)	Endocellulase, exocellulase (cellobiohydrolase), $\beta$ - glucosidase	(Qin et al. <sup>9</sup> 2016; Stamogiannou et al. 2021)
Hydroxybenzoic acid	Cocoa powder	Endocellulase, exocellulase (cellobiohydrolase), $\beta$ - glucosidase	(Sorrenti et al. 2020)

Catechin	Tea, Cocoa pod husk, Cocoa powder, <i>Mucuna</i> <i>pruriens</i> seed	$\alpha$ -amylase	(Lu et al. 2018; Rai et al. 2017; Sorrenti et al. 2020; Sun et al. 2016)
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140 **Pretreatment Techniques to Improve Enzymatic Hydrolysis Process**

141

142 There are several pretreatment strategies available to optimize the oligosaccharides  
 143 yield from lignocellulose biomass conversion. Because of its complex structure, the  
 144 crystallinity of cellulose, and porosity, lignocellulose is highly resistant to  
 145 enzymatic hydrolysis (Zoghiami and Paës 2019). Various inhibitors, such as lignin  
 146 and other polyphenolic compounds, must be treated in the pretreatment step before  
 147 entering the enzyme hydrolysis phase (Qin et al. 2016; Zoghiami and Paës 2019).  
 148 The objective of the pretreatment process is to break down the crystalline structure  
 149 of cellulose and minimize the degree of polymerization, eliminating all lignin and  
 150 breaking down hemicelluloses (Baruah et al. 2018). Proper pretreatment strategies  
 151 are required to obtain the greatest yield of oligosaccharides from the bioconversion  
 152 of different lignocellulosic substrates (Table 2).

153 The pretreatment process can result in the removal of lignin and the  
 154 destruction of the microcrystalline structure of cellulose, as well as the formation  
 155 of oligosaccharides and monosaccharides. For example, liquid hot water  
 156 pretreatment generates sugar and oligosaccharides, but the concentration of the  
 157 sugar is quite low (Huang et al. 2017). The hemicellulose concentration was  
 158 drastically reduced, although only minimal alterations in cellulose and lignin were  
 159 identified (Antczak et al. 2022). Alkali pretreatment can significantly reduce lignin  
 160 while producing fewer unwanted inhibitors. It also consumes little energy.  
 161 However, there remains a possibility that this pretreatment will result in  
 162 unrecovered salt (Jinyu Tan et al. 2021). Acid pretreatment also can disrupt the  
 163 lignocellulose and amorphous cellulose structures. Unfortunately, in contrast with  
 164 alkali pretreatment, this pretreatment consumes a large amount of energy and

165 generates a large number of byproducts (Solarte-toro et al. 2019; Jinyu Tan et al.  
 166 2021). Nowadays, research using greener technology using organic solvents such  
 167 as ionic liquid (IL) (choline, pyridine, chloride, acetate, and imidazole) and deep  
 168 eutectic solvent (DES) that are more favorable with enzymes and microbes.  
 169 However, these approaches still face challenges because solvent recovery of iLs  
 170 will cost more than other solvents while DESs have disadvantages such as high  
 171 viscosity and toxicity (Jinyu Tan et al. 2021; Zhao et al. 2022). To improve the  
 172 effectiveness of lignocellulosic waste pretreatment, a combination of pretreatment  
 173 techniques is often used.

174

175 **Table 2.** Pretreatment Techniques for Different Lignocellulosic Biomass

Biomaterials Substrate	Pretreatment Strategy	Efficiency / Yield	Ref
Rice Straw	<i>Acid:</i> Ammonia 20.93%; reaction time 48 h; temperature 42.74°C	± 13.91 g/L of fermentable glucose (± 87.24%)	(Kim et al. 2013)
Barley Straw	<i>Combination:</i> 2 g/100 g DM xylanase and polyethylene glycol (PEG) 4000	glucose yield 86.9%, xylose yield 70.2%, and acetone-butanol- ethanol (ABE) yield 135 g/kg pretreated straw	(M. Yang et al. 2015)
Sugarcane Bagasse	<i>Alkali:</i> NaOH (0.4M; t: 7 min)	Lignin was reduced from 31.71% to 12.07%. The difference between raw bagasse and processed bagasse is 19.64%.	(Z. Zhu et al. 2016)

	<p><i>Combination:</i> 22.75–38.84% of Dilute NaOH 1-3% saccharification / + Commercial conversion Enzyme Primafast 200</p>	52 (Thite and Nerurkar 2019)
	<p><i>Combination:</i> 15.6% of Alkali + <i>B. safensis</i> saccharification / M35 conversion</p>	(Thite and Nerurkar 2019)
Sugarcane Straw	<p><i>Steam Explosion:</i> Soluble XOS T: 200 °C; P: 15 yields &gt; 35 % bar; t: 10 min (w/w) and fermentable glucose yields &gt; ~78</p>	(B. Brenellia 23 et al. 2022)
Sago Pith	<p><i>Enzymatic:</i> 0.25 71.36% (w/w mL Cellulose sago pith, db) Enzyme Complex (618 CMC U/g and 139 PNPG U/g); Substrate Concentration 10% (w/w); t: 24 hours</p>	(Pinyo et al. 2016)
Sago Pith Waste (Hampas)	<p><i>Enzymatic:</i> 71.4 67.0 g/L U/g Dextrozyme fermentable sugar amylase and 20 FPU/g Acremonium cellulase; Substrate Concentration 0.09 g/mL</p>	(Husin et al. 2019)
	<p><i>Combination:</i> 43.8% glucose microwave and 40.5% hydrothermal ethanol yield hydrolysis</p>	(Thangavelu 35 et al. 2014)

Eucalyptus chip	accelerated by carbon dioxide <i>Combination:</i> instant controlled pressure drop (DIC) (controlled steam pressure (up to 7 bar) with heat (up to 170 °C) during a short time)	87% (g reducing sugar/100 g pretreated biomass)	(Messaoudi et al. 2015)
Aleppo pine cone	<i>Combination:</i> instant controlled pressure drop (DIC) (controlled steam pressure (> 7 bar) with heat (up to 170 °C) during a short time)	74% (g reducing sugar/100 g pretreated biomass)	(Messaoudi et al. 2015)
Opium poppy waste stalks	<i>Combination:</i> Combination of DIC (P: 5 bar; t: 540s) and alkaline extraction (KOH concentration 22.17%; t: 7h; V: 53.28 mL)	26.23% hemicellulose extraction yield	(Kocabas et al. 2020)
Bamboo	<i>Combination:</i> steam explosion and green-liquor (Na <sub>2</sub> S + Na <sub>2</sub> CO <sub>3</sub> )	Yields: hexoses 100.0% (% cellulose), ethanol yield of 40.1% (% cellulose)	(Gao et al. 2021)
Poplar	<i>Combination:</i> hydrothermal pretreatment (HP)	6.64% of XOS, 46.8% of fermentable	(J. Zhu et al. 2022)

	and acid hydrotropic pretreatment (AHP)	sugars (44.2% <sup>38</sup> of glucose and 2.6% of xylose), and 10.35% of Lignin nanoparticles.	
Acacia Wood	<i>Combination:</i> lime (calcium hydroxide) treatment (LT) (T: 70.9°C, t: 23.5) and hydrothermal treatment (HT) (T: 200°C; t: 10 min) <sup>27</sup>	Glucose yields 73.5%	(Lee and Yu 2021)
Durian Peel	<i>Acid:</i> 2.75% H <sub>2</sub> SO <sub>4</sub> (T: 127.14°C; t: 74.13 min) <sup>27</sup>	53 % reducing sugars	(Panakkal et al. 2021)
Corn Stover	<i>Combination:</i> ball mill-assisted alkaline peroxide pretreatment	Yields: 69.65% <sup>10</sup> XOS, 20.55% xylose, 68.94% glucose, and 21.15% gluco-oligosaccharides	(Zhang et al. 2021)

176

177

178 **Potential Microbial Thermostable Enzymes Source for Lignocellulose**

179 **Bioconversion**

180

181 Enzymatic conversion of lignocellulosic biomass, particularly the cellulose  
 182 fraction, is typically carried out at 40-50°C, which is considered a slow hydrolysis  
 183 reaction. The hydrolysis process has a low sugar yield, partial hydrolysis, and a high  
 184 risk of microbial contamination (Patel et al. 2019). These limitations could be  
 185 overcome by using thermophilic or thermotolerant microbes to produce  
 186 thermostable enzymes for lignocellulosic bioconversion (Table 3). Thermostable

187 enzymes have the advantage of being able to be stored at room temperature for  
 188 longer periods than standard enzymes. Furthermore, the high-temperature process  
 189 should avoid microbial contamination during hydrolysis and reduce the risk of  
 190 contamination and enzyme activity loss during processing. Thermostable enzymes  
 191 also enable more stable, rapid, and efficient reactions (Giovannoni et al. 2020;  
 192 Vasconcellos et al. 2015).

193 For instance, Endoglucanase (EG) can be produced by *Aspergillus terreus*  
 194 RWY and *Thermobifidia fusca UPMC 901*, with optimal active conditions of 50-  
 195 60°C and 4.0-6.0 (R. Sharma et al. 2014; Zainudin et al. 2019). In the same  
 196 environment, *Aspergillus terreus* RWY can produce cellobiohydrolase (CBH) and  
 197 xylanases (R. Sharma et al. 2014). Xylanases from *Aspergillus fumigatus* JCM  
 198 10253 are stable at 50°C for 144 hours. *Aspergillus fumigatus* JCM 10253 and  
 199 *Geobacillus* sp. HTA426 also show cellulase activity at 50-60°C (Saroj et al. 2018).  
 200 Endoglucanase and endoxylosidase activities are essential for cello-  
 201 oligosaccharides (COS) and xylooligosaccharides (XOS) production (Barbarosa et  
 202 al. 2020; Cano et al. 2020).

203

204 Table 3. Potential Microorganisms for the Production of Thermostable Enzymes

N	Microbes	Ferment ation Conditio n	Enzymes	Enzyme Activity	Enzyme Optimum Condition		Ref.
					pH	Tem perat ure	
1	<i>Aspergillus terreus</i> RWY	Solid State; 45°C for optimal production of enzyme	<i>Cellulases</i> and <i>Xylanases</i>	<i>Cellulases</i> (Filter paper cellulase (FP) (11.3±0.65 U/g-ds), endoglucanase (EG) (103±6.4	4.0-6.0	50-60°C	(R. Sharma et al. 2014)

U/g-ds),  $\beta$ -  
 glucosidase  
 (BGL)  
 (122.5 $\pm$ 8.7  
 U/g-ds),  
 cellobiohy  
 drolase  
 (CBH)  
 (10.3 $\pm$ 0.66  
 U/g-ds)),  
*Xylanases*  
 (xylanase  
 (872 $\pm$ 22.5  
 U/g-ds),  $\beta$ -  
 xylosidase  
 (22.1 $\pm$ 0.75  
 U/g-ds),  $\alpha$ -  
 L-arabino-  
 furanosidas  
 e  
 (126.4 $\pm$ 8.4  
 U/g-ds)  
 and xylan  
 esterase  
 (907 $\pm$ 15.5  
 U/g-ds)

4	<i>Geobacil lus</i> sp. HTA426	Isolated from hot spring district; optimal enzyme producti on at	<i>Cellulase</i>	CMCase activity: 103.67 U/mL	7.0	Opti mum 60°C ; Stabl e for 5 hour	(Potprom manee et al. 2017)
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		60°C for 72 h				s incu batio n at rang e 50- 70°C
4	<i>Aspergillus fumigatus</i> JCM 10253	Isolated from Warangal district, India. Solid-State Fermentation (SSF) at 50°C for 7 days	<i>Extracellular Lignocellolytic enzymes isolate</i>	CMCase (26.2 IU/mL), FPase (18.2 IU/mL), $\beta$ -glucosidase (0.87 IU/mL), and xylanase (2.6 IU/mL) (incubation time of 144 h at 50 °C)	-	60°C for crude cellulase and 50°C for FPase, $\beta$ -glucosidase and xylanase (Saroj et al. 2018)
5	<i>Thermobifida fusca</i> UPMC 901	Isolated from composted oil palm empty fruit; ferment ed on	<i>Endoglucanase</i>	CMCase: 0.9 U/mL (pH 5; T: 60 °C);	5.0	Thermal stability at 70°C (t: 24 hours)
						and (Zainudin et al. 2019)

modified  
Tryptic Soy medium  
at 50°C  
for 24  
hours

50-  
60°C  
(t:  
144  
hour  
s)

205

206

207 **Multi-Step Process and Enzymatic Hydrolysis Condition to Produce Cello-**  
208 **and Xylo-oligosaccharides**

209

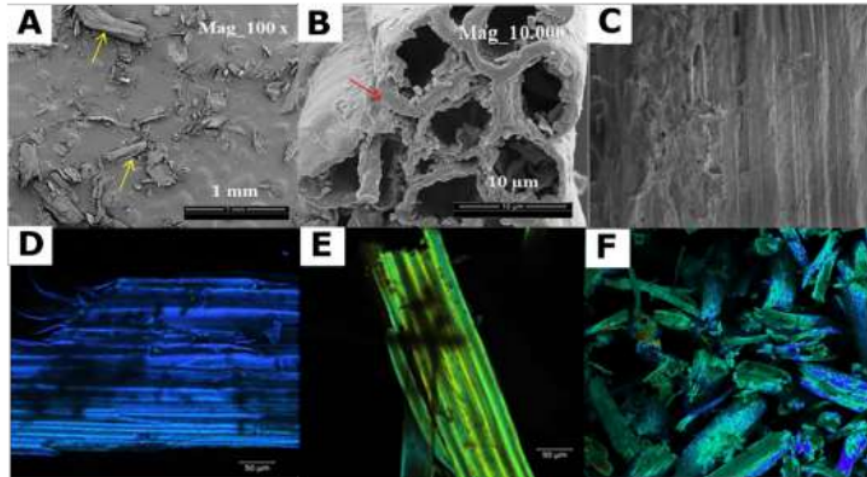
210 Enzymatic hydrolysis of lignocellulosic material can produce useful compounds  
211 like xylo- and cello-oligosaccharides. Both of these items have several health  
212 benefits, including prebiotic and anti-diabetic properties (Lin et al. 2016; J. Yang et  
213 al. 2015; Zhong et al. 2020). The production of cello- and xylo-oligosaccharides  
214 from lignocellulosic biomass can be carried out by multi-step enzymatic hydrolysis.  
215 Firstly, the pretreatment can be applied to the lignocellulosic biomass to destroy the  
216 lignin stealth, minimize the crystallinity of cellulose, and also break down the  
217 hemicellulose (Baruah et al. 2018).

218 In pretreatment I, hydrothermal pretreatment can be used to break down the  
219 crystalline structure of the lignocellulosic complex and to degrade hemicellulose.  
220 There has been no significant reduction in total lignin at this point, but there has  
221 been a change in the soluble and insoluble lignin fractions. At this point, xylo-  
222 oligosaccharides (XOS) are also being produced.

223 Marcondes et al. (2020) carried out hydrothermal pretreatment under  
224 optimum conditions at 182°C for 5.5 minutes in the absence of an external catalyst  
225 to produce ±43.61% xylo-oligosaccharides (XOS) and ±2.26% xylose. The XOS  
226 and xylose produce from this step accumulated in the liquid fraction while the solid  
227 fraction still consists of cellulose, xylan, and lignin.

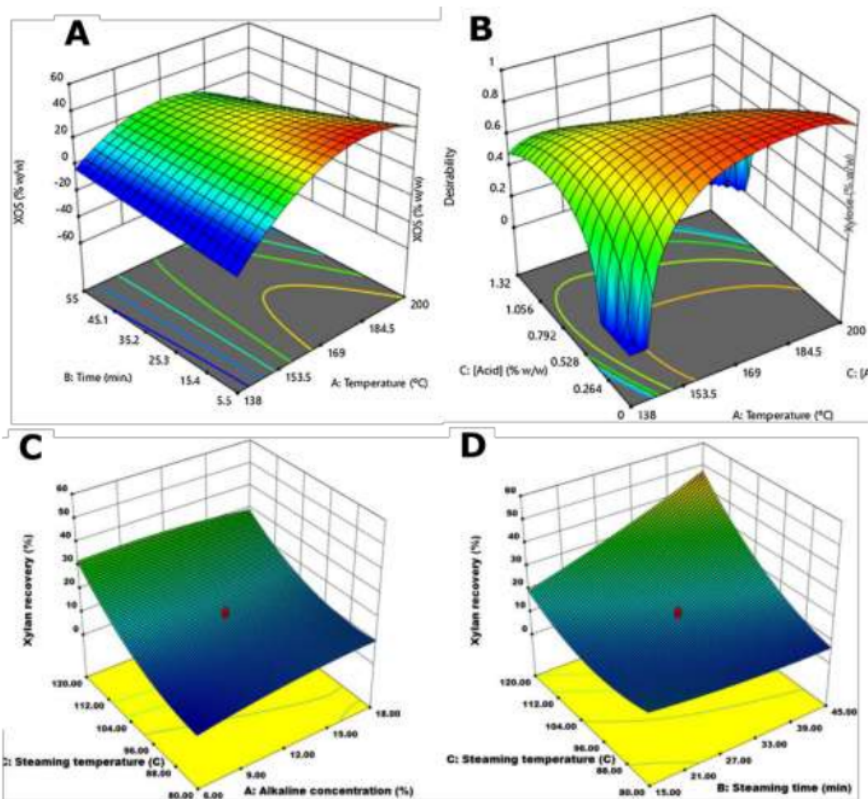
228 Espirito Santo et al. (2018) observed physical changes in the lignocellulosic  
229 structure during hydrothermal pretreatment. The hydrothermal pretreatment  
230 separated the fibers on the bundle surface from the others, but there was no

231 significant difference in total lignin. Images from Field emission scanning electron  
 232 microscopy (FESEM) and Confocal laser scanning microscopy (CLSM) show that  
 233 lignin is redistributed and that no lignin accumulates on the surface. This is due to  
 234 variations in the soluble and insoluble lignin fractions (Figure 1).



235  
 236 Figure 1. SEM imaging of (a) untreated lignocellulose structure, (b) hydrothermal  
 237 pretreatment, fibers become more separated from the surface of the fiber bundle,  
 238 (c) alkaline pretreatment, lignin dissolved and pores increased; CLSM imaging of  
 239 (d) untreated lignocellulose structure, (e) hydrothermal pretreated, and (f) alkaline  
 240 pretreated lignocellulose (Ávila-lara et al. 2015; Espírito Santo et al. 2018).

241  
 242 After the treatment, the solid phase should enter the delignification process to  
 243 remove the lignin and obtain cellulosic pulp (CP) with a small amount of lignin.  
 244 This process will optimize the enzymatic hydrolysis because of the decreasing of  
 245 inhibitors (Qin et al. 2016; Zoghlami and Paës 2019). An alkaline pretreatment can  
 246 be chosen to remove a substantial amount of lignin while avoiding the formation of  
 247 undesirable inhibitor byproducts (Martín et al. 2022; Jinyu Tan et al. 2021). Using  
 248 a combination of alkaline sulphonation and steam pretreatment, Chu et al.  
 249 previously removed around 69.37% lignin. Khat et al. (2018) discovered that the  
 250 optimal alkaline pretreatment conditions for recovering XOS were alkaline  
 251 concentration, temperature, and steaming time of 12-18%, 110-120°C, and 37.5-40  
 252 minutes, respectively. This release greatly boosts cellulose accessibility (Ávila-lara  
 253 et al. 2015; Chu et al. 2018; Yuansah et al. 2019) (Figure 2).



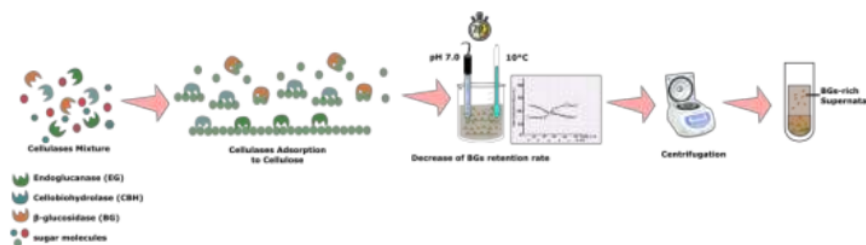
254  
 255 Figure 2. Hydrothermal pretreatment response surface for XOS yields (a) in the  
 256 absence of acid, (b) for XOS maximum and xylose minimum yields; The alkaline  
 257 pretreatment response surface (c) the effect of steam temperature and NaOH  
 258 concentration, (d) steaming temperature and time (Khat et al. 2018; Marcondes et  
 259 al. 2020).

260  
 261 Delignified cellulosic pulp can be hydrolyzed optimally by thermostable  
 262 cellulases. Using thermostable cellulase enzymes to improve the process carried out  
 263 under higher hydrolysis temperature might result in enhanced performance, such as  
 264 lower enzyme concentration and shorter hydrolysis time. A higher temperature  
 265 process is potentially lower hydrolysis charges and also more stable for longer  
 266 hydrolysis time. Endoglucanase isolated from *Thermobifidia fusca* UPMC901 has  
 267 been reported can maintain its activity without loss at temperature 50-60°C for 144  
 268 hours (Zainudin et al. 2019).

269 The thermostable cellulases used in the process should be separated from the  
 270  $\beta$ -glucosidases (BG) leaving only the endoglucanases (EG) and cellobiohydrolases

271 (CBH) acting on the substrate.  $\beta$ -glucosidase is effectively separated from other  
 272 cellulases by carrying out the “adsorption-separation” method. In this method, the  
 273 process was carried out at temperature 10°C and pH 7.0 to improve the retention of  
 274 CMCase and decrease the retention of  $\beta$ -glucosidase. The decrease in the  $\beta$ -  
 275 glucosidase retention rate is due to a change in pH and charge. EGs and CBHs have  
 276 different isoelectric points (pI), while BGs have a pI of 8.7. When the pH is adjusted  
 277 close to the pI of the BGs, the retention rate decreases. In addition, the BGs separate  
 278 and accumulate in the supernatant during centrifugation. When  $\beta$ -glucosidase is  
 279 already separated from cellulases, the process can be performed with elevated  
 280 temperatures up to 50-60°C to optimize the endoglucanase activity. The hydrolysis  
 281 performs with three-stage, improving the total COS yield accumulated in each step  
 282 by approximately 51.78% with hydrolysis approximately 75.56% (Chu et al. 2014)  
 283 (Figure 3).

284



285

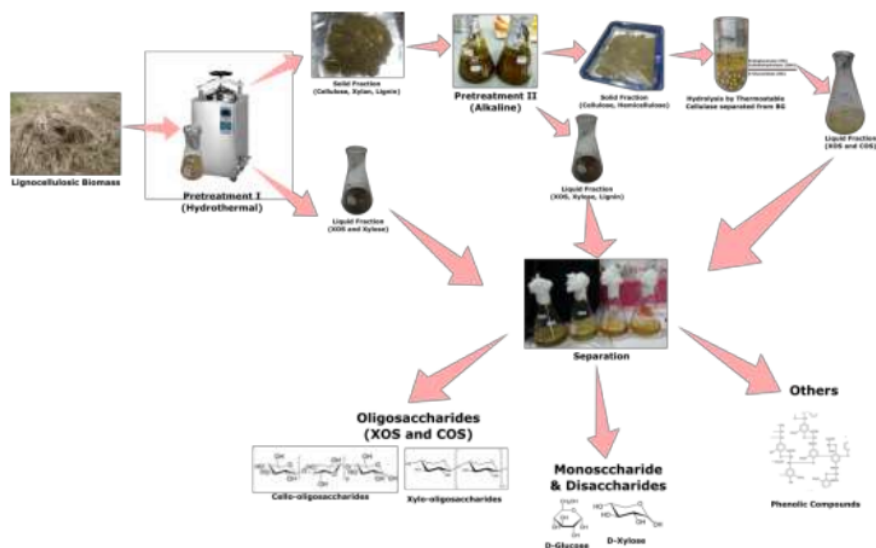
286 Figure 3. Schematic diagram of adsorption-separation method developed by (Chu  
 287 et al. 2014).

288

289 This multi-step process using thermostable enzymes has the potential to  
 290 improve the yield of COS and XOS production, decrease the use of enzyme  
 291 concentration, quick hydrolysis time, and eliminate the risk of contamination.  
 292 Furthermore, the gradual separation of products in the system in a multi-step  
 293 process minimizes the inhibition of enzyme action due to product accumulation,  
 294 allowing for a higher total yield of the desired product (Chu et al. 2014; Patel et al.  
 295 2019). Chu et al. (2014) developed a strategy for COS production using a cellulase  
 296 enzyme deficient BG by performing the "adsorption-separation" method and multi-  
 297 step enzymatic hydrolysis repeated three times. The product from each stage is  
 298 separated from the process to avoid enzyme inhibition. The hydrolysis scheme  
 299 gradually hydrolyzed the substrate over 6 hours, 6 hours, and 12 hours, yielding  
 300 COS of 20.40%, 15.53%, and 15.85%, respectively, for a total yield of 51.78%.

301 Because a large portion of COS will be produced through enzymatic  
 302 processes, cellulase inhibitors such as xylan and its derivatives must be separated  
 303 at each stage to achieve optimal yields. This product can be made using a multi-  
 304 step enzymatic hydrolysis process. Unlike COS, XOS will accumulate during  
 305 pretreatment stages I, II, and the first stage of enzymatic hydrolysis (Figure 4).  
 306 Product separation at each stage of multi-step enzymatic hydrolysis keeps XOS  
 307 from being hydrolyzed further to the monosaccharide xylose. Several studies from  
 308 Hao et al. (2022) and Zhang et al. (2021) used xylanase-cellulase enzymes in multi-  
 309 step enzymatic hydrolysis. According to the findings, XOS only detected in the first  
 310 step of enzymatic hydrolysis. Further hydrolysis revealed only monosaccharide  
 311 accumulation (Table 4).

312 For the separation process, all of the soluble oligosaccharide-rich products  
 313 from the multiple stages were collected and thin layer chromatography (TLC) can  
 314 be used to separate COS, COS, sugars (disaccharides and monosaccharides), and  
 315 other compounds (phenolic compounds and lignin derivatives) (Figure 4).  
 316  
 317



318  
 319 Figure 4. Schematic diagram of multi-step enzymatic hydrolysis strategy for COS  
 320 and XOS production  
 321  
 322

323 Table 4. Comparison of XOS and COS production yields from multi-step  
 324 enzymatic hydrolysis strategies

Source	Enzymes	Strategies	Product	Hydrolysis Yields (%)			Total Yield (%)	Ref.
				1 <sup>st</sup> stage	2 <sup>nd</sup> stage	3 <sup>rd</sup> stage		
				Corn cob	Cellulase deficient BGs	6+6+12 h		
	Xylanase - Cellulase	24+72 h	XOS	27.8	nd	n/a	27.8	(Hao et al. 2022)
Corn stover	Xylanase - Cellulase	8 +72 h	XOS	69.65	nd	n/a	69.56	(Zhang et al. 2021)

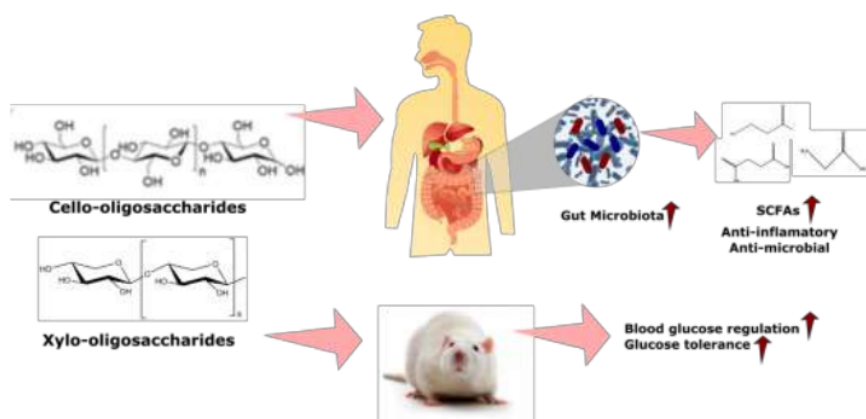
325 n/a: not applicable / not identified / not observed

326 nd: not detected

327 **Bioactivities of Cello- and Xylo-Oligosaccharides**

328

329 Cello-oligosaccharides (COS) can act as prebiotics by promoting the growth of  
 330 good microbiota in the gut, such as *Clostridium butyricum*, *Lactococcus lactis*,  
 331 *Lactobacillus paracasei*, and *Lactobacillus rhamnosus*, whereas xylo-  
 332 oligosaccharides (XOS) stimulate the bacterial groups Bifidobacilli and  
 333 Lactobacilli (Khat-udomkiri et al. 2020; Zhong et al. 2020). COS and XOS  
 334 fermentation by this group of beneficial bacteria can increase the production of  
 335 short-chain fatty acids (SCFA), which play an important anti-inflammatory role by  
 336 increasing immune cell chemotaxis and acting as an antimicrobial by disrupting  
 337 osmotic and pH balance (Jian Tan et al. 2014). XOS also has anti-diabetic activity  
 338 in type 2 diabetic rats by regulating blood glucose levels and increasing glucose  
 339 tolerance (Khat-udomkiri et al. 2020) (Figure 5).



340

341 **Figure 5. Bioactivities potential of COS and XOS**

342

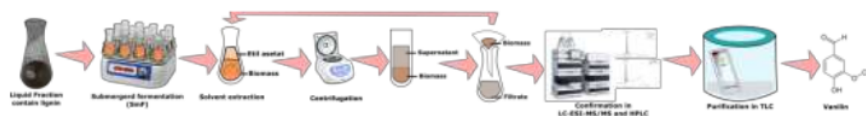
### 343 **Challenges and Future Work**

#### 344 **Pretreatment by-product Recovery During Production of XOS and COS**

345

346 In the production process of XOS and COS, lignocellulose releases several phenolic  
 347 compounds. Phenolic compounds released during the pretreatment process can be  
 348 lignin derivatives or plant secondary metabolites. Vanillin is one of the valuable  
 349 phenolics formed from lignin compounds that are discarded during the COS and  
 350 XOS manufacturing processes. To generate the vanillin from the lignin, the liquid  
 351 fraction from the pretreatment II process (Figure 4) was collected and fermented in  
 352 an orbital shaker using submerged fermentation (SmF). The vanillin compound was  
 353 extracted in ethyl acetate for 160 minutes from the fermented products. The vanillin  
 354 was separated from other phenolic compounds, then centrifuged and filtered. The  
 355 presence of vanillin in the filtrate and residue can be confirmed and quantified using  
 356 LC-ESI-MS/MS. Thin layer chromatography (TLC) was used for further  
 357 purification (Harshvardhan et al. 2017; Nurika et al. 2020) (Figure 6).

358



359 **Figure 6. Schematic diagram of vanillin recovery from post-pretreatment liquid**  
 360 **fraction in XOS and COS production**

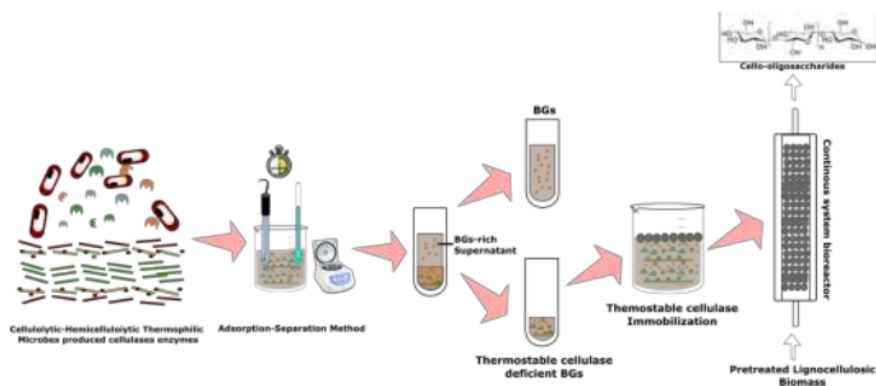
361

362 **Separation of  $\beta$ -glucosidase Enzymes Using 'Separation-Adsorption' Method**  
363 **in High Temperature and Continuous System for COS Production**

364

365 To increase the amount of COS produced, the BG enzyme must be separated  
366 from the other cellulase enzymes (endoglucanase (EG) and cellobiohydrolase  
367 (CBH)). Separating enzymes generally has a high production cost, so the  
368 adsorption-separation method developed by (Chu et al. 2014) is an alternative to  
369 separating BG from EG and CBH. The BG enzyme is separated by utilizing its  
370 isoelectric point, and decreasing the temperature results in a decrease in the  
371 retention rate of BG. When centrifuged, BG will leave the solid-liquid system and  
372 accumulate in the supernatant, while the solid residue consists of EG, CBH, and a  
373 trace of BG. Each time the product is separated from the substrate, the BG remains  
374 are gradually released in the multi-step enzymatic hydrolysis system.

375 To increase production capacity on a larger scale, thermostable cellulases can  
376 be induced from thermophilic microbes and used in the process (Yuansah et al.  
377 2019). An "adsorption-separation" method can be used to separate thermostable  
378 cellulase enzymes from BGs. The thermostable cellulase-deficient BGs can be  
379 immobilized in a matrix for continuous system bioreactors (Chu et al. 2014; Pino  
380 et al. 2018) (Figure 7). The stability of immobilized thermostable enzymes will be  
381 an advantage in large-scale production, particularly in avoiding contamination and  
382 optimizing production yields. Furthermore, the enzymes used in the process can be  
383 reused, and the enzymes do not mix with the product or require further separation.



384

385 **Figure 7. Schematic diagram of adsorption-separation methods to produce**  
386 **thermostable cellulase deficient BGs and its application in continuous system**  
387 **bioreactor**

388

389 **Conclusions**

390

391 <sup>2</sup>Based on the discussion thus far, it is possible to conclude <sup>13</sup>that lignocellulosic  
392 biomass <sup>is</sup> one of the natural sources that can be converted into a variety of  
393 functional compounds such as XOS and COS. As previously stated, because <sup>the</sup>  
394 <sup>complex structure of</sup> lignocellulose <sup>makes it difficult to</sup> convert enzymatically and  
395 results in a slow reaction and low yield, it is critical to understand the inhibitors on  
396 the lignocellulosic source to determine strategies for carrying out enzymatic  
397 hydrolysis for COS and XOS production. Thus, it is envisaged that additional  
398 studies will be conducted to establish a more efficient technique for the production  
399 of valuable products from lignocellulose.

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406 **Declaration**

407

408 <sup>8</sup>**Ethics approval and consent to participate** not applicable

409

410 **Consent for publication** not applicable

411

412 **Availability of data and materials** not applicable

413

414 **Competing Interest** <sup>1</sup>The authors declare that they have no competing interests

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418

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421 Carmando Yuansah. <sup>2</sup>The first draft of the manuscript was written by Sunrixon

422 Carmando Yuansah, Amran Laga and Pirman reviewed the article. The <sup>1</sup> authors read  
423 and approved the final manuscript.

424

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426

427

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