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Crystal structure of peroxiredoxin 3 from *Vibrio vulnificus* and its implications for scavenging peroxides and nitric oxide

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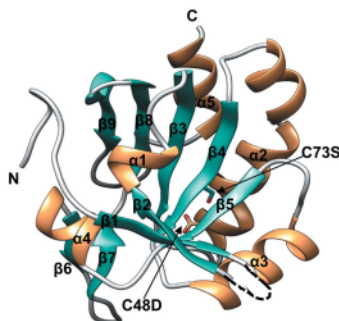
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¹⁰²Peroxi-redoxins (Prxs) are ubiquitous cysteine-based peroxidase enzymes. Recently, a new type of Prx, VvPrx3, was identified in the pathogenic bacterium *Vibrio vulnificus* as being important for survival in macrophages. It employs only one catalytic cysteine residue⁷⁷ to decompose peroxides. Here, crystal structures of VvPrx3 representing its reduced and oxidized states have been determined, together with an H₂O₂-bound structure, at high resolution. The crystal structure representing the reduced Prx3 showed a typical dimeric interface, called the ⁹³type interface. However, VvPrx3 forms an oligomeric interface mediated by a disulfide bond between two catalytic cysteine residues from two adjacent dimers, which differs from the doughnut-like oligomers that appear in most Prxs. Subsequent biochemical studies showed that this disulfide bond was induced by treatment with nitric oxide (NO) as well as with peroxides. Consistently, NO treatment induced expression of the *prx3* gene in *V. vulnificus*, and VvPrx3 was crucial for the survival of bacteria in the presence of NO. Taken together, the function and mechanism of VvPrx3 in scavenging peroxides and NO stress via oligomerization are proposed. These findings contribute to the understanding of the diverse functions of Prxs during pathogenic processes at the molecular level.

1. Introduction

Most organisms have evolved efficient defence systems that rapidly detoxify potentially damaging reactive oxygen species (ROS). ROS are generated during aerobic respiration and in the host immune system (Simples *et al.*, 2005; Cabiscol *et al.*, 2000; Halliwell, 2006). Peroxi-redoxins (Prxs) represent a superfamily of thiol-specific antioxidant peroxidases that use an invariant cysteine residue(s) (Engelman *et al.*, 2013). They are highly conserved throughout evolution and are ubiquitously found from humans to bacteria. Prxs decompose peroxides such as hydrogen peroxide (H₂O₂) and alkyl hydroperoxides in the cytosol (Rhee *et al.*, 2005; Wood *et al.*, 2003; Engelman *et al.*, 2013). Recent research has reported that Prxs are also involved in the detoxification of reactive nitrogen species (RNS), which are produced in macrophages as a key component for killing invading bacteria (Wong *et al.*, 2002; Bryk *et al.*, 2000). Prxs have been found to detoxify peroxynitrite (ONOO⁻) and nitrosylated thiols (Fang, 2004; Beckman & Koppenol, 1996; Pfeiffer *et al.*, 1997).

Prxs have a peroxidatic cysteine (C_P) that reacts with peroxides in the first step of catalysis. The C_P is oxidized to cysteine sulfenic acid (C_P-SOH) by attacking the O—O bond of peroxides during the peroxidase reaction, resulting in the decomposition of peroxides (Wood *et al.*, 2003). The C_P-SOH is then reduced during the resolution step, which is needed to

recycle the protein. Depending on the resolution step, Prxs are classified into three types, referred to as typical 2-Cys, atypical 2-Cys and 1-Cys Prxs (Hall *et al.*, 2011). Both types of 2-Cys Prx have an additional conserved cysteine residue called the resolving cysteine (C_R), which forms a stable intermolecular or intramolecular disulfide bond with the C_P -SOH. While C_R is located in a separate subunit in typical 2-Cys Prxs, it is located within the same subunit in atypical 2-Cys Prxs (Seo *et al.*, 2000; Hall *et al.*, 2009). The disulfide bond in 2-Cys Prxs can be broken into two free thiols by cellular reductants (Chae *et al.*, 1999). In contrast, the C_P -SOH of 1-Cys Prxs has been implicated to be directly reduced by cellular reductants such as glutathione (Grx) and/or the reduced form of glutathione (GSH) (Noguera-Mazon *et al.*, 2006; Kang *et al.*, 1998; Rhee *et al.*, 2005). In a yeast 1-Cys Prx, the C_P -SOH has been proposed to form a disulfide bond with the C_P residue of another molecule of Prx (Pedrajas *et al.*, 2016).

Although the oligomeric states of Prxs differ depending on the type of Prx, most oligomeric Prxs are formed via two major distinct interfaces: B-type and A-type interfaces (Parsonage *et al.*, 2005). In typical 2-Cys Prxs, dimers are primarily formed via the B-type interface, in which C_P and C_R are located in different protomers and disulfide formation between C_P and C_R is facilitated by peroxides (Echalier *et al.*, 2002). In most 2-Cys Prxs, five dimers are further associated via the A-type interface to form a doughnut-like decamer (Wood *et al.*, 2002; Parsonage *et al.*, 2005). Most atypical 2-Cys Prxs and 1-Cys Prxs exhibit dimeric assembly via the A-type interface (Evrard *et al.*, 2004; Li *et al.*, 2005). Exceptionally, a non-A and non-B oligomeric interface between C_P and C_R was observed in human Prx5 (*HsPrx5*), which belongs to the atypical 2-Cys Prx family (Smeets *et al.*, 2008) and showed redox-dependent structural changes in the C_P -containing region. The structural change was distinguished from that in some typical 2-Cys Prxs (Wood *et al.*, 2003; Parsonage *et al.*, 2005). In the presence of excess peroxides the C_P can be overoxidized to cysteine sulfinic acid or cysteine sulfonic acid (C_P -SO₂H or C_P -SO₃H, respectively). Interestingly, the protomers with overoxidized C_P exhibit the same conformation as in Prxs in the reduced state (Mizohata *et al.*, 2005; Nakamura *et al.*, 2006).

Vibrio vulnificus is a highly virulent foodborne pathogen that can cause fatal septicaemia (Oliver, 2005; Horseman & Surani, 2011). The bacterium has three different types of Prx: Prx1, Prx2 and Prx3. Prx1 and Prx2 belong to the typical 2-Cys Prx family of proteins and their expression is controlled by the H₂O₂-activated transcription factor OxyR (Kim *et al.*, 2014; Jo *et al.*, 2015). *V. vulnificus* Prx3 (*VvPrx3*) belongs to the 1-Cys class of Prxs, with one conserved catalytic cysteine (Cys48), and has been shown to be involved in pathogenicity in a mouse model. According to the PeroxiRedoxin Classification Index (PREX) database (<http://www.csb.wfu.edu/prex>), it belongs to the Prx5 subfamily. *VvPrx3* expression is controlled by IPR, which is activated by oxidative stress and iron starvation (Lim *et al.*, 2014). In this study, we determined crystal structures of *VvPrx3* in two different forms and in complex with H₂O₂. Based on the crystal structures, we present a molecular mechanism of 1-Cys Prxs that highlights the oligomer interface

containing a disulfide bond between two peroxidatic cysteine residues. More importantly, we examined the expression of *prx3* in *V. vulnificus* induced by nitric oxide gas (NO); consistently, *VvPrx3* reacts with nitric oxide.

2. Experimental procedures

2.1. Plasmid construction

The *VvPrx3* (C48D/C73S) and *VvPrx3* (C73S) genes of *V. vulnificus* were cloned into the pET-21c vector (Invitrogen). To clone the *VvPrx3* gene for the expression of *VvPrx3* (C73S), two primers (forward, 5'-GGTCATATGATCGCTC AAGGCCAAACTTTACC-3'; reverse, 5'-CCTCTCGAGCG CGCAAGAATCGTTTCAGC-3') were designed, which contained NdeI and XhoI sites (underlined), respectively. Site-directed mutagenesis was performed in two subsequent PCR reactions for the expression of *VvPrx3* (C48D/C73S) (Ho *et al.*, 1989). The PCR products and pET-21c plasmid were digested by NdeI and XhoI, and the digested products were ligated with DNA ligase. The ligation products were transformed into *Escherichia coli* C43 (DE3) cells by heat shock. Finally, the recombinant plasmid was confirmed by DNA sequencing.

2.2. Expression and purification of recombinant proteins

Recombinant proteins were expressed in LB medium containing 50 µg ml⁻¹ ampicillin, with growth at 310 K to an OD₆₀₀ of 0.5. Samples were induced with 0.5 mM isopropyl β-D-1-thiogalactopyranoside (IPTG) for 6 h at 303 K. The cells were harvested and resuspended in lysis buffer consisting of 20 mM Tris-HCl pH 8.0, 0.15 M NaCl, 2 mM β-mercaptoethanol. After disrupting the cells with a French press, the cell debris was removed by centrifugation. The supernatant was loaded onto Ni-NTA affinity resin (Qiagen, The Netherlands) that had been pre-incubated with lysis buffer. The target protein was eluted with lysis buffer supplemented with 250 mM imidazole. Eluted recombinant proteins were diluted threefold with 20 mM Tris-HCl pH 8.0 buffer containing 2 mM β-mercaptoethanol and loaded onto a HiTrap Q column (GE Healthcare, USA). A linear gradient of increasing NaCl concentration was applied to the column. The fractions containing the protein were pooled, concentrated and applied onto a size-exclusion chromatography column (HiLoad 16/600 Superdex 200 pg, GE Healthcare) pre-equilibrated with lysis buffer. The final protein samples were concentrated to 20 mg ml⁻¹ using a centrifugal filter concentration device (Millipore; 10 kDa cutoff) and stored frozen at 193 K until use.

2.3. Crystallization, structure determination and refinement

Crystallization was performed at 287 K using the hanging-drop vapour-diffusion method. Crystals of Prx3 (C48D/C73S) were obtained using a precipitant solution consisting of 0.6 M NaCl, 0.8 M sodium citrate, 0.1 M Tris-HCl pH 6.5. The structure of reduced Prx3 (C48D/C73S) was determined using the molecular-replacement method with *MOLREP* in the *CCP4* package (Winn *et al.*, 2011) using a putative thioredoxin

reductase from *Burkholderia cenocepacia* (PDB entry 4f82; Seattle Structural Genomics Center for Infectious Disease, unpublished work) as a search model. The crystal of Prx3 (C48D/C73S) obtained under reduced conditions [reduced Prx3 (C48D/C73S)] belonged to space group $P3_221$, with unit-cell parameters $a = 73.9$, $b = 73.9$, $c = 62.3$ Å and one protein molecule in the asymmetric unit (Table 1). After 5 d, Prx3 (C73S) crystals appeared with precipitant solution consisting of 0.1 M sodium citrate pH 5.5, 0.2 M ammonium acetate, 22% (w/v) PEG 4000. Crystals were exchanged into the appropriate mother liquor containing 20% (v/v) glycerol, mounted on cryoloops and flash-cooled in a liquid-nitrogen stream at 100 K. X-ray diffraction data sets were collected on beamline 5C at the Pohang Accelerator Laboratory and were processed with the *HKL-2000* package (Otwinowski & Minor, 1997). The crystal of Prx3 (C73S) obtained under oxidized conditions [oxidized Prx3 (C73S)] belonged to space group $P2_12_12_1$, with unit-cell parameters $a = 39.5$, $b = 57.4$, $c = 124$ Å and two protein molecules in the asymmetric unit (Table 1). The oxidized Prx3 (C73S) crystals were flash-cooled using a crystallization solution with the same cryoprotectant as used for the reduced Prx3 (C48D/C73S) crystals in a nitrogen stream at 100 K. The initial model of oxidized Prx3 (C73S) was determined by molecular replacement using the structure of reduced Prx3 (C48D/C73S) as a model and was refined at 2.0 Å resolution (Table 1). The final structure of oxidized Prx3 (C73S) was refined at 1.48 Å resolution using *PHENIX* (Afonine *et al.*, 2012). To obtain H₂O₂-bound Prx3 (C48D/C73S) crystals, 20 mM H₂O₂ was added to the reservoir solution and 500 µM H₂O₂ was added to the hanging-drop solution containing the crystals. After 10 d, the H₂O₂-bound Prx3 (C48D/C73S) crystals were flash-cooled using the same cryoprotectant as used for the reduced Prx3 (C48D/C73S) crystals in a nitrogen stream at 100 K. The crystals belonged to space group $P1$, with unit-cell parameters $a = 75.08$, $b = 97.72$, $c = 97.49$ Å and 12 protein molecules in the asymmetric unit. Statistics regarding data collection and processing are presented in Table 1.

2.4. Oxidation of VvPrx3 by H₂O₂ or NO

To prepare the reduced VvPrx3 (C73S) protein, 10 mM DTT was added to the purified VvPrx3 (C73S) protein and was then removed using a HiPrep 26/10 Desalting column (GE Healthcare, USA) pre-equilibrated with 20 mM Tris-HCl pH 8.0 buffer containing 150 mM NaCl. The reduced VvPrx3 (C73S) mutant protein was reacted with various concentra-

Table 1
Data-collection and refinement statistics.

Values in parentheses are for the highest resolution shell.

	Oxidized Prx3 (C73S)	Reduced Prx3 (C48D/C73S)	H ₂ O ₂ -bound Prx3 (C48D/C73S)
Data collection			
Space group	$P2_12_12_1$	$P3_221$	$P1$
a, b, c (Å)	39.5, 57.4, 124	73.9, 73.9, 62.3	75.1, 97.7, 97.5
α, β, γ (°)	90, 90, 90	90, 90, 120	78.7, 67.3, 67.3
Resolution (Å)	50–1.48 (1.51–1.48)	50–1.90 (1.93–1.90)	50–1.90 (1.93–1.90)
R_{merge}	0.07 (0.28)	0.06 (0.39)	0.06 (0.36)
$\langle I/\sigma(I) \rangle$	25.9 (4.96)	39.0 (4.57)	14.5 (2.07)
Completeness (%)	98.6 (95.1)	99.3 (99.5)	90.0 (82.6)
Multiplicity	10.0 (6.0)	13.3 (7.1)	3.4 (2.2)
Refinement			
Resolution (Å)	18.9–1.48 (1.52–1.48)	32.0–1.90 (1.96–1.89)	19.7–1.91 (1.93–1.91)
No. of reflections	47559	14790	137002
$R_{\text{work}}/R_{\text{free}}$	0.181/0.198	0.220/0.258	0.195/0.244
Total No. of atoms	2568	1189	14956
No. of ligands	0	0	4
No. of water molecules	231	37	805
Wilson B factor (Å ²)	12.1	15.7	17.3
R.m.s. deviations			
Bond lengths (Å)	0.007	0.002	0.003
Bond angles (°)	1.1	0.57	0.57
Ramachandran plot			
Favoured (%)	95.5	94.7	93.4
Allowed (%)	4.5	5.3	6.5
Outliers (%)	0.0	0.0	0.1
Molecules in asymmetric unit	2	1	12
PDB code	5k2i	5k1g	5k2j

The R_{free} value was calculated using a test-set size of 9.95%.

tions of H₂O₂ or tert-butyl hydrogen peroxide (*t*-BOOH) 30 min at 298 K in 1 ml 20 mM Tris-HCl buffer pH 8.0 containing 150 mM NaCl. The reaction mixture was treated with iodoacetamide (IAA) to stop the reaction and subjected to SDS-PAGE under nonreducing or reducing conditions. To analyze Prx3 oxidation during the time course, samples were treated with 50 µM H₂O₂ for 0–60 s and the reaction was quenched with 10 mM *N*-ethylmaleimide (NEM). For reaction with NO, reduced VvPrx3 (C73S) protein (10 nmol) was treated with 1 mg NO-releasing PLC-PEI nanoparticles (NO/PPNPs) in 10 ml reaction mixture in 20 mM Tris pH 7.5, 150 mM NaCl. The amount of NO was calculated based on the release of 3.3 nmol NO per hour by 1 mg NO/PPNPs (Nurhasni *et al.*, 2015).

2.5. Competitive kinetics with horseradish peroxidase (HRP)

To prepare the reduced VvPrx3 (C73S) protein, the purified protein was treated with 10 mM DTT for 30 min at room temperature. After reduction of the protein, residual DTT was removed using a HiPrep 26/10 Desalting (GE Healthcare, USA) column pre-equilibrated with 20 mM Tris-HCl pH 7.5 buffer containing 150 mM NaCl. Reaction mixtures containing 5 µM horseradish peroxidase (HRP; Sigma-Aldrich) and various concentrations (0–9.15 µM) of reduced VvPrx3 (C73S) were treated with 2.5 µM H₂O₂ at room temperature in reaction buffer consisting of 50 mM sodium phosphate pH 7.4, 150 mM NaCl. The concentration of HRP was calculated using the absorbance at 403 nm ($\epsilon_{403} = 1.02 \times 10^5 \text{ M}^{-1} \text{ cm}^{-1}$) and

the concentration of VvPrx3 (C73S) was determined at 280 nm ($\epsilon_{VvPrx3} = 8.48 \times 10^3 \text{ M}^{-1} \text{ cm}^{-1}$). The competitive kinetics of VvPrx3 (C73S) were determined using a previously described procedure (Oguscu *et al.*, 2007; Winterbourn & Peskin, 2016; Cox *et al.*, 2009). In brief, the ratio of inhibition of HRP oxidation $[F/(1 - F)]$ was measured at 403 nm with a spectrophotometer using a 10 mm cuvette. The second-order rate constants (k_1) of VvPrx3 (C73S) were determined from the slope of a plot of $[F/(1 - F)]k_{HRP}[HRP]$ against $[VvPrx3 (C73S)]$ ($k_{HRP} = 1.7 \times 10^7 \text{ M}^{-1} \text{ s}^{-1}$).

2.6. Gel-filtration chromatography

A Superose 6 HR 10/30 column coupled to an FPLC instrument (Biologic Duo Flow, Bio-Rad) was equilibrated with 20 mM Tris-HCl pH 8.0, 150 mM NaCl. Samples were

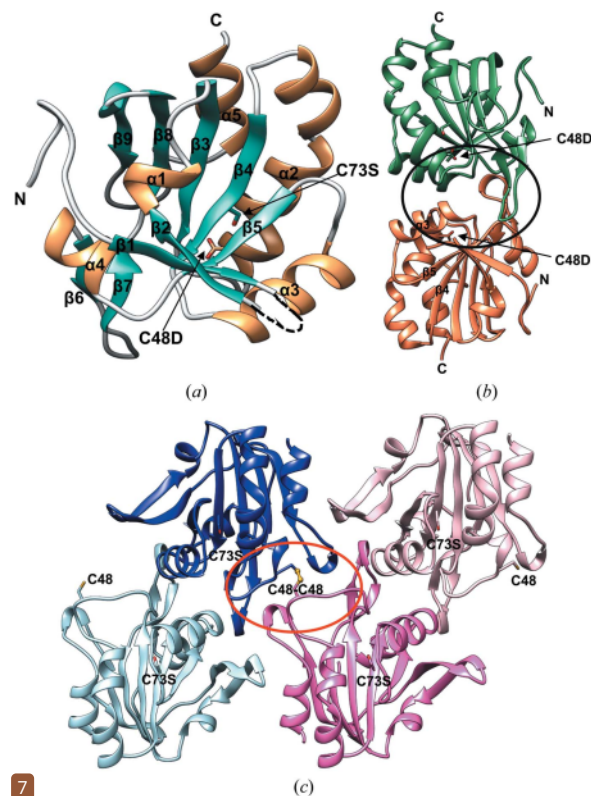


Figure 1 Overall structure and oligomeric interfaces of VvPrx3. (a) Ribbon diagram of the reduced Prx3 (C48D/C73S) monomer. (b) Dimeric structure of reduced Prx3 (C48D/C73S) formed by an A-type interface. (c) The asymmetric unit of oxidized Prx3 (C73S) with intact Cys48. Two Cys48 residues from adjacent protomers form a disulfide bond. The crystals of oxidized Prx3 (C73S) were grown with a reducing agent. Four protomers are drawn in different colours and the side chains of Cys48 and Ser73 are displayed as ball-and-stick models. The red circle represents the C-type interface.

incubated with 10–30 μM H_2O_2 for 30–60 min at room temperature before loading them onto the column to form the intermolecular disulfide bonds of VvPrx3 (C73S). Samples were subjected to chromatography, which was performed at a flow rate of 0.2 ml min^{-1} . Several standard preparations were run to calibrate the column: ferritin (440 kDa), aldolase (158 kDa), conalbumin (75 kDa), ovalbumin (43 kDa), carbonic anhydrase (29 kDa) and ribonuclease A (13.7 kDa) (Supplementary Fig. S8). The absorbance at 280 nm was used to monitor the presence of the proteins.

2.7. Survival in the presence of nitric oxide (NO)

The *V. vulnificus* cultures were grown to an A_{600} of 0.5 in LBS broth. B broth supplemented with 2.0% (w/v) NaCl at 30°C and the cells were harvested by centrifugation at 1500g for 7 min at room temperature. The cell pellets were resuspended in M9 minimal medium supplemented with 0.4% (w/v) glucose. The bacterial cells [$\sim 4 \times 10^7$ colony-forming units (CFU)] were exposed to 0.1% (w/v) NO/PPNPs (Nurhasni *et al.*, 2015) for 2 h and an aliquot was acquired every 20 min. The number of live bacterial cells in the aliquots was determined as the CFU on LBS agar plates. Bacterial strains and plasmids are described in Supplementary Table S1.

2.8. RNA purification and transcript analysis

Wild-type *V. vulnificus* cells grown to an A_{600} of 0.5 in LBS broth were exposed to various concentrations of PLGA-PEI nanoparticles (PPNPs) and NO/PPNPs for 20 min and were harvested to isolate the total RNA. Total RNAs were isolated using an RNeasy Mini kit (Qiagen). For quantitative real-time PCR (qRT-PCR), the concentration of total RNA from the strains was measured using a NanoVue Plus spectrophotometer (GE Healthcare). The cDNA was synthesized from 1 μg total RNA using an iScript cDNA-synthesis kit (Bio-Rad). Real-time PCR amplification of cDNA was performed using a Chromo 4 real-time PCR detection system (Bio-Rad) with a pair of specific primers (Supplementary Table S2) as described previously (Jang *et al.*, 2016). The relative expression levels of *iscR* and *prx3* mRNA in the same amount of total RNA were calculated using the 16S rRNA expression level as an internal reference for normalization.

3. Results

3.1. Structural determination of reduced Prx3 (C48D/C73S), oxidized Prx3 (C73S) and H_2O_2 -bound Prx3 (C48D/C73S)

VvPrx3 contains two cysteine residues: Cys48 and Cys73. Cys48 is the C_p residue that is crucial for the function of the protein. However, Cys73 is not involved in protein function because the enzymatic activity of VvPrx3 (C73S) was similar to that of wild-type VvPrx3 in the GSH/Grx3/GR system (Lim *et al.*, 2014). To focus on the function of Cys48 and prevent unwanted disulfide-bond formation at Cys73 during SDS-PAGE analyses, Cys73 was replaced with a serine residue in this study. To investigate an oxidized structure of Prx3 containing a disulfide bond, the crystal of the oxidized VvPrx3

(C73S) protein was obtained under crystallization conditions without a reducing agent. The crystal structure of VvPrx3 (C73S) [reduced oxidized Prx (C73S) in this study] was determined at 1.5 Å resolution in space group $P2_12_12_1$ using the molecular replacement method. The resulting structure was refined to an R_{work} of 0.181 and an R_{free} of 0.198 (Table 1). The final model comprises residues 1–157 and contains two protein molecules in the asymmetric unit (Table 1). To study the structure of Prx3 in the reduced state, crystals of a VvPrx3 (C48D/C73S) variant were also obtained. The C_P Cys48 in VvPrx3 was replaced with a non-oxidizable Asp residue, which could also partly mimic an over-oxidized form (Cys-SO₂H) of VvPrx3 (Jo *et al.*, 2015). The crystal belonged to a different space group, $P3_221$, and the VvPrx3 (C48D/C73S) structure [named reduced Prx (C48D/C73S) in this study] was determined by the molecular replacement approach. The final model, refined at 1.9 Å resolution ($R_{\text{work}} = 0.220$ and $R_{\text{free}} = 0.258$), contains one protomer (residues 1–18 and residues 22–157) in the asymmetric unit (Table 1). To examine the interactions between VvPrx3 and peroxides, the H₂O₂-bound Prx3 (C48D/C73S) structure was determined at 1.9 Å resolution. A high concentration (500 μM) of H₂O₂ was incubated with crystals of reduced Prx3 (C48D/C73S), which changed the space group and unit-cell parameters. The resulting structure was refined to an R_{work} of 0.195 and an R_{free} of 0.244. Electron-density maps were well defined for all 12 protomers in the asymmetric unit (Table 1).

All three crystal structures of VvPrx3 revealed a protomer exhibiting a compact and spherical structure without a C-terminal extension region, similar to the previously reported 1-Cys Prx AhpE from *Mycobacterium tuberculosis* (Supplementary Figs. S1 and S2; Li *et al.*, 2005). The VvPrx3 protomer is formed based on a central five-strand β-sheet (β5–β4–β3–β8–β9; Fig. 1a). α-Helices α2 and α5 are located on one side of the central β-sheet, while α-helix α4 and the β-hairpin (β1 and β2) are on the other side of the central

β-sheet. Strands β4–β3–β8–β9 and helices α2, α4 and α5 comprise the typical Trx fold. In the peripheral region, a β4–α3–β5 structural motif is located in the A-type dimeric interface (Figs. 1a and 1b). C_P Cys48 (or the mutated Asp48) is located at the N-terminus of the long, kinked α2, similar to other Prxs. The mutated Ser73 is buried in the Trx fold region (Fig. 1a). Structural superposition of reduced Prx (C48D/C73S) on human PrxV (PDB entry 1oc3; sequence identity 47%; Evrard *et al.*, 2004) revealed high structural similarity, with an r.m.s.d. of 1.46 Å between the C^α positions of 247 matched amino acids (Supplementary Fig. S2).

3.2. Comparison of structures of VvPrx3

All three crystal structures of VvPrx3 [reduced Prx3 (C48D/C73S), oxidized Prx3 (C73S) and H₂O₂-bound Prx3 (C48D/C73S)] showed a similar dimeric assembly in the crystals characterized by an A-type dimeric interface, similar to that in AhpE from *M. tuberculosis* (Supplementary Figs. S1 and S2; Li *et al.*, 2005). In the reduced Prx3 (C48D/C73S) structure the C_P-containing α2 helix at the active site adopted the ‘fully folded’ or reduced conformation usually observed in Prx structures in the reduced and overoxidized states (Fig. 2; Mizohata *et al.*, 2005; Schröder *et al.*, 2000).

Strikingly, a dimeric interface was found in the oxidized Prx3 (C73S) that was distinct from both the A-type and B-type interfaces (Figs. 1a and 1b). This dimeric interface was similar to an exceptional oligomeric interface observed in the oxidized structure of HsPrx5, although this structure contains the disulfide bond between C_P and C_R (Smeets *et al.*, 2008), while VvPrx3 directly forms a dimeric interface through the intermolecular disulfide bond between the C_P Cys48 residues. This interface was designated the ‘C-type’ interface in this study since these oligomeric interfaces was induced by the disulfide bonds.

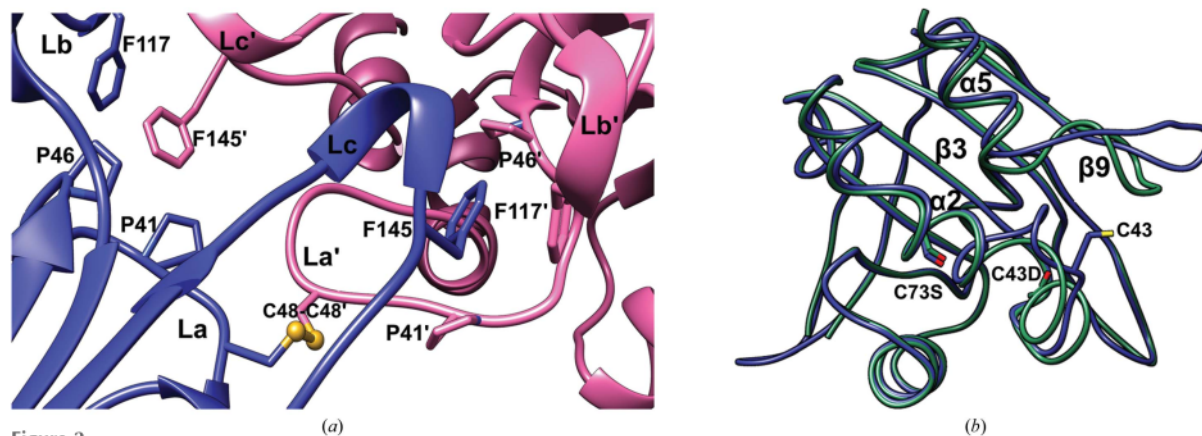


Figure 2
Structural comparison of VvPrx3 in reduced and oxidized states. (a) A magnified view focusing on the interactions at the C-type interface (Fig. 1c; red circle) with the intermolecular disulfide bond present between Cys48 residues. The side chains involved in the hydrophobic core are shown in ball-and-stick representation. The loops in this interface are labelled La, Lb and Lc, indicating the loop connecting β3 and α2, the loop connecting β6 and β7, the loop connecting β9 and α5, respectively. Primes are used in the labels to distinguish the protomers. (b) Structural superposition of VvPrx3 in the reduced form (green) and the disulfide-bridged (oxidized) form (blue).

The C-type interface mainly consists of two C_P-containing α2 helices from adjacent protomers (Fig. 1c). Hydrophobic interactions were observed in the C-type interface that appear to further stabilize this dimeric interface. The Phe145 residue in loop β9–α5 of one protomer forms a hydrophobic core with Phe117 in loop β6–β7, while Pro41 and Pro46 form a hydrophobic core in the loop connecting β3 and α2 of the other molecule (Fig. 2a). Owing to disulfide bonds and hydrophobic interactions, a significant conformational alteration was observed in the active site of helix α2 containing C_P and nearby loops. In addition, loop β9–α5 moved outwards from the core and loop β3–α2 became ordered in the oxidized Prx3 (C73S) structure representing the oxidized structure, compared with the reduced Prx3 (C48D/C73S) structure representing the overoxidized or reduced conformation (Fig. 2b). In *HsPrx3*, the conformational change of loop β7–α6 responding to loop β9–α5 of *VvPrx3* was induced by forming an intramolecular disulfide bond depending on the redox state (Supplementary Fig. S3).

3.3. Intermolecular disulfide formation of *VvPrx3* reacted with peroxides, H₂O₂ and *t*-BOOH

To confirm whether the observed disulfide bond in the crystal structure is generated in the solution state, we

performed the following biochemical experiment with H₂O₂ and *t*-BOOH, which are known substrates of *VvPrx3* (Dubbs & Mongkolsuk, 2007; Beckman & Koppenol, 1996; Lim *et al.*, 2014). Reduced *VvPrx3* (C73S) protein was incubated with H₂O₂ or *t*-BOOH and analyzed by SDS–PAGE under non-reducing conditions to observe intermolecular disulfide-bond formation. While boiling the protein sample for SDS–PAGE, free cysteine residues appeared to form nonspecific disulfide bonds. To avoid these nonspecific disulfide bonds, the free thiol-specific alkylating agent iodoacetamide was added to the sample prior to boiling to block free cysteines. We observed that the protein bands were up-shifted to a dimeric size on the SDS–polyacrylamide gels on treatment with H₂O₂ or *t*-BOOH (Figs. 3a and 3b). The protein bands shifted back to a monomeric size when a reducing agent was added to cleave disulfide bonds during boiling. These results indicate that two *VvPrx3* protomers were linked by an intermolecular disulfide bond between two Cys48 residues. Interestingly, the intensities of the dimeric protein bands decreased at higher concentrations of peroxide. Maximum band intensities were shown at 10 μM H₂O₂ or *t*-BOOH (Figs. 3a and 3b), likely owing to over-oxidation of C_P at high peroxide concentrations (Claiborne *et al.*, 1999; Rhee *et al.*, 2007). These findings indicate that the intermolecular disulfide bond between C_P residues is induced by the peroxides H₂O₂ and *t*-BOOH, unlike the prevailing

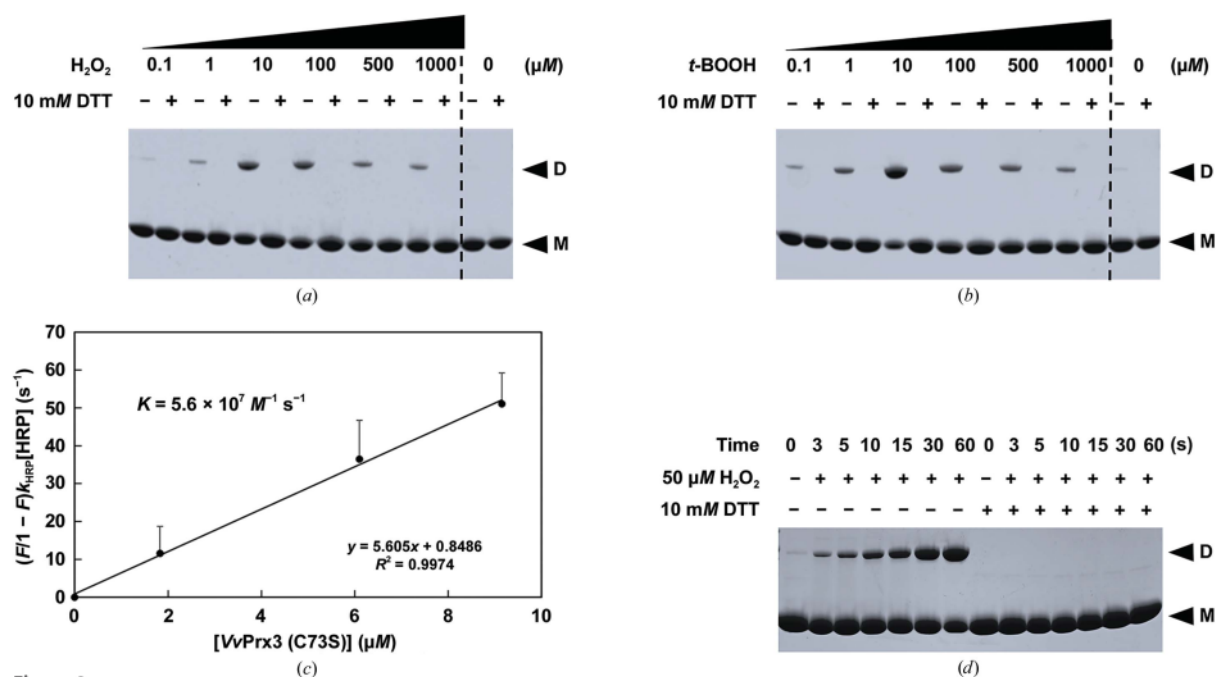


Figure 3 Intermolecular disulfide-bond formation in *VvPrx3* (C73S) on treatment with peroxide. Proteins were treated with H₂O₂ (a) or *tert*-butyl hydroperoxide (*t*-BOOH) (b) at the concentrations shown for 30 min and the reactions were stopped by adding iodoacetamide. Samples were boiled in the presence (+) or absence (–) of 10 mM DTT for SDS–PAGE analysis. (c) Determination of the second-order reaction rate constant for the reaction of *VvPrx3* and H₂O₂ using competition kinetics with HRP. The fitting line was calculated by the least-squares method and the error bars reflect the standard deviation from three independent experiments. (d) Purified *VvPrx3* (C73S) (70 μM) was reacted with 50 μM H₂O₂ for the given times and analyzed by SDS–PAGE. At a time of 30 s, ~30 μM protein was linked by a disulfide bond as estimated using the density-measuring program *ImageJ*. The protein band for disulfide bond-containing *VvPrx3* (D) and bands that lack a disulfide bond (M) are indicated.

mechanisms for 1-Cys Prxs via the direct reduction of C_P-SOH by Grx or GSH (Noguera-Mazon *et al.*, 2006; Kang *et al.*, 1998; Rhee *et al.*, 2005).

The second-order reaction rate constant (k_1) for H₂O₂ consumption by VvPrx3 was measured by employing competition kinetics with horseradish peroxidase (HRP; Winterbourn & Peskin, 2016; Cox *et al.*, 2009; Ogusucu *et al.*, 2007). The measured value was $5.6 \times 10^7 \text{ M}^{-1} \text{ s}^{-1}$, which was of the same order as that for human Prx2 (Peskin *et al.*, 2007; Fig. 3c). This finding indicates that VvPrx3 is as efficient as 2-Cys Prxs. To estimate the reaction rate of dimerization of VvPrx3, the amount of dimeric VvPrx3 was measured using the same SDS-PAGE method with different incubation times of VvPrx3 and H₂O₂. We observed that almost half of the VvPrx3 formed an intermolecular disulfide bond within 30 s (Fig. 3d), indicating that the reaction velocity for disulfide-bond formation between VvPrx3-SOH and VvPrx3-SH is comparable to a similar reaction with human Prx2 or Prx3 (Gupta & Carroll, 2014). Taken together, our findings suggest that dimerization of VvPrx3 could provide a rapid kinetic path to resolving C_P and is central to the catalysis of the enzyme.

3.4. Redox-dependent oligomerization of VvPrx3

A protein oligomer representing the oxidized state was found in the oxidized Prx3 (C73S) crystals (Fig. 1c). The

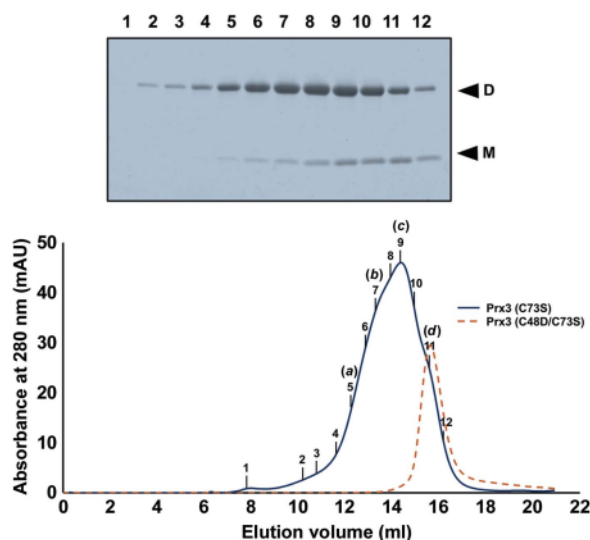


Figure 4
Oligomerization states of VvPrx3 in solution were analyzed using size-exclusion chromatography, and fractions were further analyzed by SDS-PAGE under nonreducing conditions. The VvPrx3 (C73S) protein (33 μM) was incubated with H₂O₂ (15 μM) for 60 min before applying size-exclusion chromatography. The result of the native gel electrophoresis is provided in Supplementary Fig. S5. The molecular size was calculated from the standard plot in Supplementary Fig. S8. The molecular sizes of the fractions labelled (a)–(d) correspond to decamer (~244 kDa), hexamer (~143 kDa), tetramer (~81 kDa) and dimer (~42 kDa), respectively. The protein band of disulfide-bond-containing VvPrx3 (D) and bands that lack a disulfide bond (M) are indicated. As a control for the non-oxidized sample, VvPrx3 (C73S/C48S) was analyzed using the same column (shown by a broken line).

oligomer is generated in a linear arrangement by alternating connections between the protomers via A-type contacts and C-type contacts (Supplementary Fig. S4). As observed in Fig. 1(c), peroxides induce a disulfide bond between protomers through the C-type interface. The A-type dimer in the reduced state is further linked by the disulfide bond between the C_P residues at the C-type interface, which can lead to the linear oligomer observed in oxidized Prx3 (C73S) crystals.

To investigate changes in the oligomeric states of VvPrx3 proteins induced by oxidative stress, the oligomeric state and disulfide-bond formation of VvPrx3 proteins in the presence/absence of H₂O₂ was analyzed by combining size-exclusion chromatography and SDS-PAGE. In these experiments, two Prx variants were used: C73S and C48D/C73S. To create a disulfide bond between the C_P residues, VvPrx3 (C73S) protein was treated with H₂O₂ before loading it onto a size-exclusion chromatography column pre-equilibrated with lysis buffer devoid of reducing agent. The elution profile showed a broad early-eluting peak and shoulders before achieving the dimeric size of VvPrx3, reflecting mixed forms of higher oligomers. Decamers, hexamers, tetramers and dimers were observed on the size-exclusion chromatography column when the molecular sizes were estimated based on the elution volume. The higher oligomeric forms in solution were further confirmed by native gel electrophoresis (Supplementary Fig. S5). The higher oligomeric forms contained more VvPrx3 dimers that are linked by a disulfide bond between two C_P residues, as judged by SDS-PAGE under nonreducing conditions (Fig. 4). These results indicate that the higher oligomeric forms of VvPrx3 were created by noncovalently associated A-type interfaces and disulfide-linked C-type interfaces, as observed in crystals of oxidized Prx3 (C73S).

3.5. H₂O₂ binding site

Ovoid-shaped electron density indicating H₂O₂ was found near Asp48 and the active-site region in the H₂O₂-bound Prx3

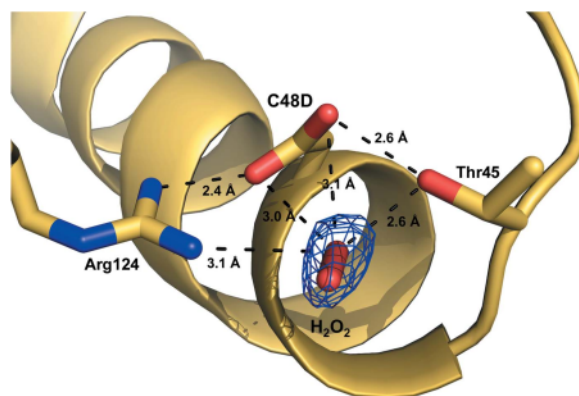


Figure 5
The H₂O₂ binding site in H₂O₂-bound Prx3 (C48D/C73S) mimics the overoxidized structure. Bound H₂O₂ and the residues involved are shown in stick representation at a resolution of 1.9 Å. Broken lines indicate the interactions between residues and H₂O₂. The ovoid-shaped electron density indicates H₂O₂.

(C48D/C73S) structure (Fig. 5 and Supplementary Fig. S6). The H₂O₂ binding site is fully accessible from the solvent since it is exposed to the external medium. The structure is analogous to that of the H₂O₂-bound archaeal 2-Cys Prx thio-redoxin peroxidase from *Aeropyrum pernix* K1 (Nakamura *et al.*, 2010), in which the corresponding cysteine residue was overoxidized and shares the H₂O₂ binding site (Supplementary Fig. S6*d*). Thus, it is likely that substrate binding near C_P is important in the rapid catalysis of VvPrx3. In Prx structures only one O atom of H₂O₂ near Cys48 (or Asp48 in our structure) makes further interactions with VvPrx3, while the other O atom makes no interaction with H₂O₂-bound Prx3 (C48D/C73S) (Fig. 5). Both O atoms in H₂O₂ were extensively involved in polar interactions in OxyR, which can recognize only H₂O₂ and not alkyl hydrogen peroxides (Jo *et al.*, 2015). This difference in H₂O₂ recognition might result in the different substrate specificities of Prx and OxyR proteins.

3.6. NO induces intermolecular disulfide bonds with further oligomerization

Increasing attention has been given to the NO-mediated reactions of host immune systems (Bogdan *et al.*, 2000; Nathan & Shiloh, 2000). A large amount of NO is synthesized in macrophages or epithelial cells stimulated by pathogens or by substances secreted by pathogens (Beckman & Koppenol, 1996; Fang, 2004). To investigate the functional relevance of VvPrx3 and NO, we examined whether NO can induce the disulfide bonding of VvPrx3, similar to H₂O₂ (Engelman *et al.*, 2013). Although Prxs are known to scavenge reactive nitrogen species such as the peroxynitrite that is rapidly generated by combining NO and superoxide ion (Fang, 2004; Beckman & Koppenol, 1996; Pfeiffer *et al.*, 1997), direct treatment with NO has not been tested on Prxs. We treated *V. vulnificus* with NO-releasing polymeric nanoparticles (NO/PPNPs), which have recently been developed to allow the sustained release of NO in solution (Nurhasni *et al.*, 2015), and then measured the mRNA levels of *prx3* and its known transcription factor *iscR*. The results showed that the mRNA levels of *prx3* and *iscR* significantly increased, implying that NO induces VvPrx3 in *V. vulnificus* via IScR (Fig. 6*a*).

We next tested whether NO directly induces disulfide bonds in the VvPrx3 protein. NO/PPNPs were treated with VvPrx3 protein, and the disulfide-bond formation of the protein was analyzed by SDS-PAGE and size-exclusion chromatography (Fig. 6*b* and Supplementary Fig. S7). Compared with air-oxidized VvPrx3 protein, significant protein bands representing the disulfide bond-linked VvPrx3 increased on the

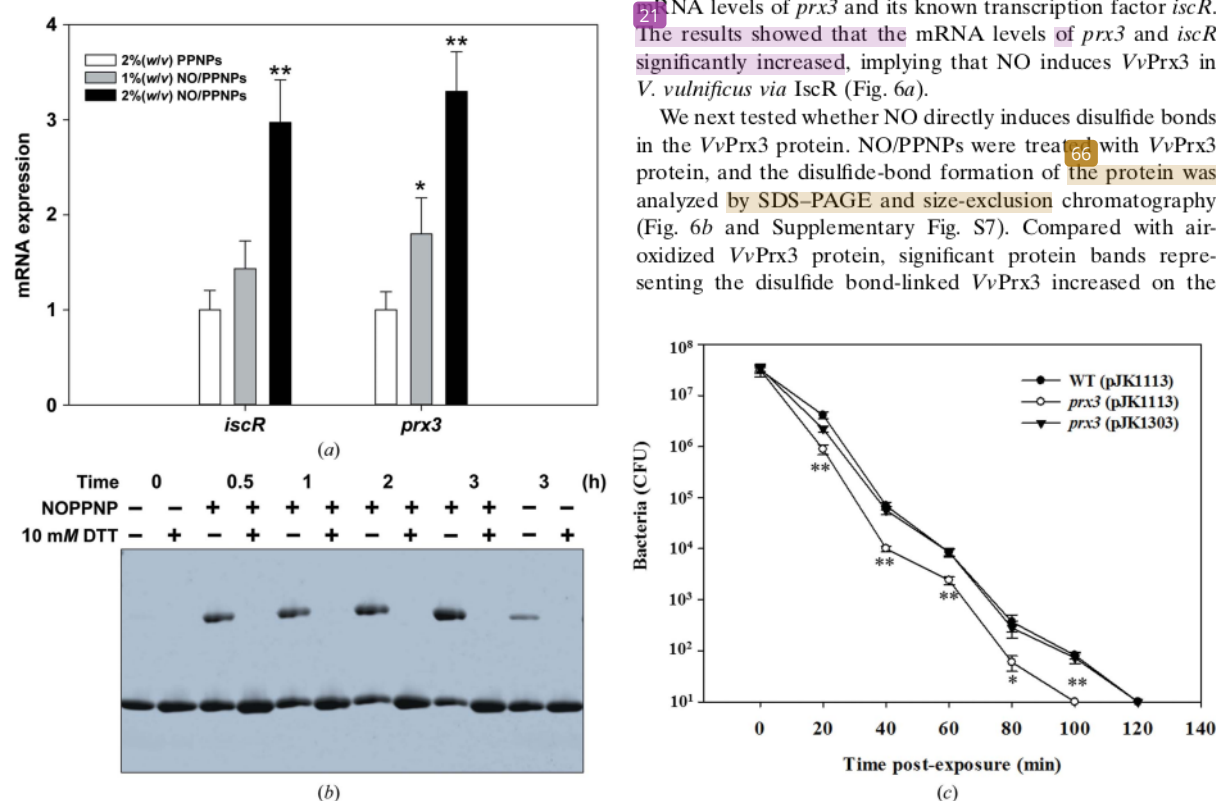


Figure 6 Effect of NO exposure on *Vvprx3* expression and VvPrx3 protein. (a) Total RNA was isolated from wild-type cells grown aerobically to an A₆₀₀ of 0.5 after exposure to various concentrations of PLGA-PEI nanoparticles (PPNPs) or NO/PPNPs for 20 min. PPNPs were used as blank nanoparticles. The *iscR* and *prx3* mRNA levels were determined by qRT-PCR analyses, and the *iscR* and *prx3* mRNA levels in the wild type exposed to 0.2% (w/v) PPNPs were set to 1. Error bars represent the standard deviation (SD). *, *P* < 0.05; **, *P* < 0.005. (b) VvPrx3 (C73S) was incubated with NO/PPNPs for the times shown. A total of 3.3 μM NO accumulated in the reaction mixture per hour. The reaction was terminated by adding 2.5 mM iodoacetamide to prevent nonspecific disulfide bonds. Aliquots were taken from the reaction mixture and loaded onto SDS-PAGE under reducing conditions (DTT +) or nonreducing conditions (DTT -). (c) Survival of *V. vulnificus* strains in the presence of NO stress. Wild-type bacteria (WT) or *prx3*-deleted bacteria (*prx3*) were transformed with the empty expression vector pJK1113 or the functional *prx3*-containing vector pJK1303. Well grown bacterial cells were exposed to NO/PPNPs for the times shown and viable cells were counted as colony-forming units (CFU) on LBS agar plates. Error bars represent the standard deviation (SD). *, *P* < 0.05; **, *P* < 0.005.

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SDS–polyacrylamide gel under nonreducing conditions after treatment of the nanoparticles releasing $\sim 1 \mu\text{M}$ NO. The NO concentration in solution is much lower than the effective concentration ($\sim 10 \mu\text{M}$) of H_2O_2 for $Vv\text{Prx3}$ (Fig. 6b). Thus, our results suggest that $Vv\text{Prx3}$ could respond to NO stress through a similar mechanism as for H_2O_2 stress.

3.7. $Vv\text{Prx3}$ supports the survival of *V. vulnificus* under NO stress

To evaluate the role of $Vv\text{Prx3}$ in resistance to NO in bacteria, the survival rates of wild-type and *prx3* gene-deleted *V. vulnificus* strains were measured. Compared with the wild-type strain, the *Prx3* mutant strain exhibited substantially impaired growth in a minimal medium containing NO/PPNPs. When the functional *prx3* gene (pJK1303) was added back to the mutant strain, the bacterial survival rate was restored (Fig. 6c). Taken together, our results suggest that $Vv\text{Prx3}$ plays an important role in scavenging NO imposed on the bacteria.

4. Discussion

Crystal structures of $Vv\text{Prx3}$, a bacterial 1-Cys Prx, were determined in different redox states in this study. The H_2O_2 binding site near the C_P residue was also shown. Although structural and sequence similarities to the 1-Cys Prx AhpE from *M. tuberculosis* were identified, especially in the dimeric assembly with the A-type interface, our structures provide unexpected and important insights into the function and molecular mechanism of Prxs. One striking feature of oxidized Prx3 (C73S) is a dimeric C-type interface with structural changes, which was previously observed in the oxidized form

of HsPrx5 belonging to the atypical 2-Cys Prx family (Smeets *et al.*, 2008). Since the C-type interface has been observed in two different kinds of Prxs, this oligomeric interface could be shared by many other Prx proteins. This study further observed that protein–protein interaction at the C-type interface was induced by both NO and peroxides. Owing to the oxidation-dependent C-type interface, $Vv\text{Prx3}$ formed a higher order linear oligomer in response to peroxides and NO. This linear oligomer's structure is distinct from the circular decameric assembly of typical 2-Cys Prxs, which is generated by alternate A-type and B-type interfaces. Since the linear oligomer of $Vv\text{Prx3}$ is generated only in the oxidized state, the redox-dependent oligomerization of $Vv\text{Prx3}$ differs from that of typical 2-Cys Prxs. Interestingly, this linear oligomeric form was also predicted in HsPrx5 (Smeets *et al.*, 2008).

According to the prevailing mechanisms, the C_P in 1-Cys Prxs is itself oxidized to $\text{C}_\text{P}\text{-SOH}$ through attack of the substrate peroxide. $\text{C}_\text{P}\text{-SOH}$ is reduced by the reductase Grx or GSH (Noguera-Mazon *et al.*, 2006; Kang *et al.*, 1998; Rhee *et al.*, 2005). This study presents another pathway for the reduction of $\text{C}_\text{P}\text{-SOH}$ in 1-Cys Prxs, which might complement the direct reduction of $\text{C}_\text{P}\text{-SOH}$ by Grx or GSH under certain circumstances. In this pathway, the catalytic cysteine residue (C_P) in 1-Cys Prx plays a dual role in the catalytic cycle. Not all $Vv\text{Prx3}$ proteins are simultaneously oxidized by H_2O_2 . Thus, remaining or unreacted $Vv\text{Prx3}$ could act on other $Vv\text{Prx3}$ s with $\text{C}_\text{P}\text{-SOH}$. The $\text{C}_\text{P}\text{-SOH}$ can be reduced and resolved by free C_P in a different dimer of $Vv\text{Prx3}$, in which the free C_P acts as a C_R . The complementary structural features at the C-type interface provide specific interactions with $Vv\text{Prx3}$ proteins, facilitating the prompt formation of a disulfide bond between $\text{C}_\text{P}\text{-SOH}$ and the free C_P of $Vv\text{Prx3}$ proteins (Fig. 1c).

This reaction could also occur successively, leading to the formation of the linear oligomers observed in the crystal structure (Supplementary Fig. S4). Finally, cellular GSH or Grx that is reduced by GSH reduces the disulfide bond in the oligomer, as proposed previously (Kim *et al.*, 2014). NO can be held in the H_2O_2 binding site of the $Vv\text{Prx3}$ active site to be readily attacked by C_P . The free C_P from another $Vv\text{Prx3}$ molecule resolves the resulting nitrosylated cysteine ($\text{C}_\text{P}\text{-NO}$), as depicted in Fig. 7. Many different members are present in the 1-Cys Prx family. Of these, 1-Cys Prxs from yeast have previously been characterized. As in $Vv\text{Prx3}$, the intermolecular disulfide bond between C_P was observed in yeast 1-Cys Prx on an SDS–polyacrylamide gel under nonreducing conditions and predicted a doughnut-like oligomer formation in the oxidized state (Pedrajas *et al.*, 2016). We speculate that yeast 1-Cys Prx might share a catalytic

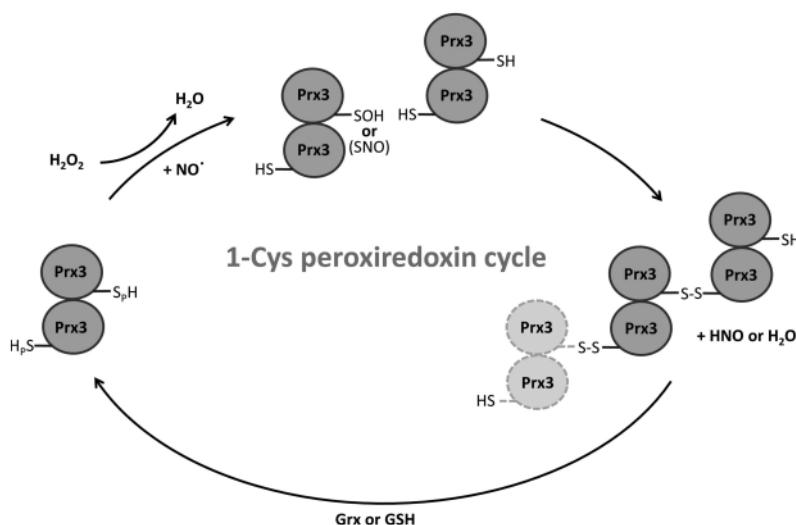


Figure 7

Proposed catalytic cycle of 1-Cys $Vv\text{Prx3}$. Each Prx3 polypeptide is represented by grey circles, with the peroxidatic Cys48 highlighted according to its redox state: sulfhydryl, $-\text{SH}$; sulfenic acid, $-\text{SOH}$; nitrosothiol, $-\text{SNO}$; disulfide, $-\text{S}-\text{S}-$. Potential conformations of additional dimers are indicated by light grey ovals.

cycle with VvPrx3, although further studies are required to elucidate its mechanism.

The pathogenic bacterium *V. vulnificus* can defend against NO stress to increase survival in this harsh host environment. A previous study showed that VvPrx3 is important in the pathogenicity of *V. vulnificus* in mice (Lim *et al.*, 2014). In this study, we observed that NO induced *prx3* together with *iscR* in *V. vulnificus* and that NO is a substrate of VvPrx3. These findings suggest that VvPrx3 is important for bacterial survival in NO-challenged environments and might account for the involvement of VvPrx3 in pathogenicity in the host.

Diverse peroxiredoxins exist in nature and play different roles in the scavenging of toxic radicals. We have proposed a molecular mechanism for bacterial I-Cys Prxs and a new role in scavenging NO in this study, which may in part account for the pathogenesis of *V. vulnificus*. Our findings expand the understanding of the molecular mechanism of survival of pathogenic bacteria in the host environment by scavenging diverse oxidative and nitrosative stresses.

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